Exam Date & Time: 30-Apr-2022 (10:00 AM - 01:00 PM)

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MANIPAL ACADEMY OF HIGHER EDUCATION

	Pharmaceutics-1 [PCE-BP1031 - S2]			
Marks: 75	Duration: 13	80 mins.		
	I Multiple Choice Questions (MCQs)			
Answer all	the questions. Section Duration:	30 mins		
1)	Which of these excipients is typically not used in liquid dosage forms.			
		(1)		
	1) Vehicle 2) Surfactants 3) Hydrocolloids 4) Binders			
^^)	20° u/p alcohol is approximately% v/v			
	1) 120 2) 46.8 3) 80 4) 68.52	(1)		
3)	Which of the following methods can prevent incompatibility due to liquefaction of substances mixed together?			
	1) Separate dispensing Mixing individual Mixing mixture of Any 2) substance with inert absorbent inert absorbent above	(1)		
4)	2 lb = grains			
	1) 480 2) 6500 3) 11520 4) 875	(1)		
5)	Which of these is a unit solid dosage form?			
	1) Tablet 2) Capsule 3) Cachet 4) All the above	(1)		
6)	How many pennyweights ares equal to 1 ounce troy?			
	1) 48 2) 20 3) 24 4) 40	(1)		
7)	Which of these dosage forms are typically not meant for external use?			
	1) Tablet 2) Lotion 3) Gargle 4) Collodion	(1)		
8)	Which of these vehicles for liquid dosage forms should never be sterile?			
	1) Purified Water for Injection 3) Bacteriostatic Water 4) None of the above	(1)		
9)	1 decagram = kg	(1)		

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	1) 0.01 2) 0.1 3) 0.001 4) 10					
10)	How much of ABC is required to prepare 100 mL of a 5% w/v ABC solution?					
	1) 5 g 2) 50 g 3) 0.5 g 4) 50 mg	(1)				
11)	This person is known as Father of Botany					
	1) Hippocrates 2) Theophrastus 3) Paracelsus 4) Galen	(1)				
12)	This book is example for non-official compendia					
	1) Indian Pharmacopoeia 2) Merck Pharmacopoeia 3) United States Pharmacopoeia 4) National Formulary	(1)				
13)	This part of prescription contains the list of ingredients with their quantities	ŕ				
	1) Superscription 2) Inscription 3) Subscription 4) Signatura	(1)				
14)	These powders composed of acids and sodium bicarbonate, when added to water, liberate CO_2					
	1) Explosive powders 2) Efflorescent powders 3) Effervescent powders 4) Deliquescent powders	(1)				
15)	First pass metabolism of drug can not be avoided by using					
	1) Nasal liquids 2) Rectal preparations 3) Oral liquids 4) Topical Preparations	(1)				
16)	Following statement is NOT TRUE for liniments					
	Liniments are 1) Aqueous or oily based solutions Liniments Liniments are applied without 2) without Liniments are used for external Use 3) are not to be applied to	(1)				
17)	friction Only broken skin					
17)	Anhydrous glycerine is added in ear drops as it					
	Permits the drug to Tends to remove None All of 1) remain in the ear for a long time 2) moisture from 3) of the above above	(1)				
18)	Following dosage form exhibit higher rate of bioavailability.					
	1) Powders 2) Solutions 3) Emulsions 4) Suspensions	(1)				
19)	This base provide occlusive effect to the ointment	(1)				

	1) Hydrocarbon Bases	2) Absorption base	Water- 3) removable bases	Water-soluble bases						
20)	The acrylic acid polymer commonly used as gel base is									
	1) Carboxy methyl cellulose	2) Carbo	omer 3) Poloxamer	r 4) Methyl cellulose	(1)					
II Long Answers										
Answer all t	the questions.									
1)	Explain physical and chemical incompatibilities in pharmaceuticals.									
2)	A) Discuss the type of suspension which is difficult to redisperse (4 Marks). B) Classify emulsifying agents with two examples each. Write the importance of emulsifying agents (6 Marks).									
		III Short An	swers							
Answer all t	he questions.									
1)	Describe five factors affect	ing selection of do	se of drug.		(5)					
2)	Give a classification of the systems of weights and measures. Convert 1 \Im , 1 \Im , 1 \Im and 1 \textmd{tb} into grams.									
3)	Describe five types of excipients in liquid dosage forms.									
4)	Explain the features of absorption bases used in ointments with example.									
5)	Explain the handling of compounding prescription.									
6)	Explain the cold compression method of preparation of suppositories with diagram.									
7)	What are the characteristic features of the gels? Classify gel bases with two examples each.									
End		ž.								

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