

Question Paper

Exam Date & Time: 27-Nov-2023 (10:00 AM - 01:00 PM)



MANIPAL ACADEMY OF HIGHER EDUCATION

Pharmaceutical Inorganic Chemistry [PCH-BP104T]

Marks: 75

Duration: 180 mins.

I Multiple Choice Questions (MCQs)

Answer all the questions.

Section Duration: 30 mins

- 1) Electrolytes replacement therapy is needed when (1)
Diarrhoea
Prolonged fever and vomiting
Heavy loss of water
All the above
- 2) The difference between unmeasured anion and unmeasured cation is known as (1)
Anionic gap
Cationic gap
Trace ionic gap
Extra electrolytes
- 3) Major extracellular cation is (1)
Calcium
Potassium
Ammonium
Sodium
- 4) ORS does not have (1)
Glucose
Sodium citrate
Sodium chloride
Sodium bicarbonate
- 5) One of the following is used as dental cement (1)
Zinc chloride
Zinc oxide
Sodium chloride
Potassium chloride
- 6) Astringents show which type of mechanism of action (1)
Styptic
Antiperspirant
Anti-irritant
All the above
- 7) About 65% of Iron in human body is present in (1)

Myoglobin

Enzyme

RBC

Ferritin

8) Beta particle is (1)

With zero mass

Negatively charged

Identical to electron

All the above

9) Which of the following is most penetrating radiation (1)

Alpha rays

Beta rays

Gama rays

Sound waves

10) In the treatment of arthritis following radiopharmaceutical is used (1)

Nitrogen-13

Samrium-153

Erbium-169

Strontium-89

11) The first Indian Pharmacopoeia's publishing started in the year 1944 under the Chairmanship of (1)

BR Chopra

TB Chopra

RN Chopra

B.N Ghosh

12) One of the solvent below is of low toxicity (1)

Cyclohexane

Acetic acid

Carbon tetrachloride

Acetonitrile

13) In method A for limit test of heavy metals IP the pH of solutions is adjusted to (1)

2-3

3-4

4-5

5-6

14) The chemical formula of the substance which is filtered off in the limit tests for chloride and sulphate (1) in potassium permanganate is

MnO

MnO₂

Mn₂O₃

Mn₂O₅

15) Yellow colour stain formed in arsenic limit test is due to the formation of (1)

Mercury arsenate

Mercury arsenide

Mercury arsenic acid

Mercury Arsenous acid

16) 0.8%w/v of _____ solution is isotonic with serum. (1)

Ammonium chloride

Sodium chloride

Hydrochloride

Silver chloride

17) The synonym of aqueous iodine is (1)

Bleaching powder

Strong tincture of Iodine

Lugol's solution

Weak iodine solution

18) Hydrogen peroxide is used as (1)

Antiseptic

Acidifying agent

Protective

Antioxidant

19) Agents that act by increasing the osmotic load of gastro intestinal tract by absorbing large quantity of water and thereby stimulate peristalsis are called as (1)

Bulk

purgatives

Saline purgatives

Stimulant purgatives

Emollient purgative

20) An inorganic agent act by irritating gastric mucosa & then stimulate respiratory tract secretion. (1)

Terpin hydrate

Ammonium chloride

Ambroxol

Sodium carbonate

II Long Answers

Answer all the questions.

1) Explain the methods to measure the tonicity of a solution. Explain the different classes of buffers based on the basis of their uses. (6+4) (10)

2) Explain the terms antiseptics, disinfectants, and germicides. Explain the preparation and assay principle of boric acid and Iodine. (10) (10)

III Short Answers

Answer all the questions.

1) How do you prepare Potash alum? Write the mechanism of action and uses of Potash alum. (5)

2) Explain Gas filled tube counters as a device to measure radioactivity. (5)

3) Highlight the important properties of Alpha radiations. What are the uses of I-131? (5)

4) Write a note on preparation of activated charcoal. (5)

5) What modification is done in the limit test for chloride for potassium permanganate and sodium salicylate? Why? (5)

6) How do the impurities get incorporated into the pharmaceutical preparations during the (5)

manufacturing process? Explain

- 7) Give the method of preparation, physical property, assay and uses of Epson salt. (5)

-----End-----