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SECOND YEAR B. PHARM. DEGREE EXAMINATION - MAY 2016

SUBJECT: PATHOPHYSIOLOGY (PPR 201T) (2014 REGULATION)

Tuesday, May 17, 2016

Time: 10:00 - 13:00 Hrs.

Max. Marks: 70

- Answer ALL the questions.
- ∠ Long answer questions:
- 1. Explain the cellular events in acute inflammation.
- 2. Classify acute renal failure based on etiopathogensis.
- 3. Define anaemia. Explain its etiopathogensis.

 $(10 \text{ marks} \times 3 = 30 \text{ marks})$

- 4. Short answer questions:
- 4A. Explain pathophysiology of atherosclerosis.
- 4B. Define diabetes. Explain its complications.
- 4C. Explain the pathophysiology of gastric ulcer.
- 4D. Explain type 2 hypersensitivity reaction.
- 4E. Explain pathophysiology of idiopathic parkinsonism.
- 4F. Explain nuclear changes in necrosis with help of neat diagram.

 $(5 \text{ marks} \times 6 = 30 \text{ marks})$

- 5. Give reasons for the following:
- 5A. Necrosis associated with inflammation.
- 5B. In premenopausal women risk of atherosclerosis is lesser compared to menopausal women.
- 5C. Primaquine use in G6PD deficiency patients leads to acute renal failure.
- 5D. H.Pylori causes duodenal ulcer.
- 5E. CD4+ cell count decreases in HIV patients.



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SECOND YEAR B. PHARM. DEGREE EXAMINATION - MAY 2016

SUBJECT: PATHOPHYSIOLOGY (PTH 201) (CREDIT BASED SYSTEM)

Tuesday, May 17, 2016

Time: 10:00 - 13:00 Hrs.

Max. Marks: 50

Answer ALL questions.

∠ Long Essay:

- 1A. Enumerate the etiological factors of ischemic cell injury.
- 1B. Explain the general mechanism by which ischemic cell injury occur.
- 1C. Explain the cellular and tissue changes that occur in ischemic injury.

(2+2+4 = 8 marks)

- 2A. Differentiate humoral and cell mediated immunity with suitable examples.
- 2B. Explain Major Histocompatibility System.
- 2C. Describe the role of antibodies in immune system.

(3+3+2 = 8 marks)

- 3A. Define acute renal failure and enumerate the etiological factors for the same.
- 3B. Explain various clinical features of acute renal failure.
- 3C. Explain complications of chronic renal failure.

(2+3+3 = 8 marks)

4. Short Essay:

- 4A. Classify anemia based on hematological tests.
- 4B. Explain the pathogenesis of parkinsonism and enumerate four clinical symptoms.
- 4C. Explain the etiopathogenesis of atherosclerosis.
- 4D. Describe pathophysiology and enumerate the clinical symptoms of chronic obstructive pulmonary disease.

 $(4 \text{ marks} \times 4 = 16 \text{ marks})$

5. Short Answer:

- 5A. Explain type I Hypersensitivity reaction.
- 5B. Differentiate metastatic and benign cancer.
- 5C. Enumerate four opportunistic infections of AIDS.
- 5D. Differentiate bronchitis and emphysema.
- 5E. Enumerate four etiological agents for chronic hepatitis.



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SECOND YEAR B. PHARM. DEGREE EXAMINATION - MAY 2016

SUBJECT: PHARMACEUTICAL MICROBIOLOGY (PBT 202T) (2014 REGULATION)

Saturday, May 07, 2016

Time: 10:00 - 13:00 Hrs.

Max. Marks: 70

Answer ALL the questions.

∠ Long answer questions:

- 1. Draw a neat labelled diagram of typical bacterial cell. Discuss in detail the features of Gram positive and Gram negative cell walls.
- 2. Draw a neat labelled diagram of an industrial autoclave and describe its design.
- 3. What is an antibiotic policy, why it is needed? Discuss briefly on various types of antibiotic policies.

 $(10 \text{ marks} \times 3 = 30 \text{ marks})$

4. Short answer questions:

- 4A. Write a note on growing anaerobic bacteria.
- 4B. Explain the effect of time of contact and temperature on the activity of disinfectants.
- 4C. Briefly outline the procedure for microbiological assay of antibiotics by two level factorial assay.
- 4D. Enlist the demerits of Phenol coefficient tests and explain any two.
- 4E. It is often necessary to amputate the affected limb in gas gangrene to save the patient. Why? How this can be avoided?
- 4F. Write the causative agent, mode of transmission, important symptoms, prevention and treatment of rabies.

 $(5 \text{ marks} \times 6 = 30 \text{ marks})$

5. Give reasons for the following:

- 5A. Ultrathin sectioning is necessary for TEM but not for SEM.
- 5B. Bacterial endospores have high heat resistance.
- 5C. Aspergillus species can be differentiated from Penicillium species microscopically.
- 5D. Autoclaving is not suitable to sterilise milk and vaccines.
- 5E. Infections from *E coli* cannot be treated with vancomycin.

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SECOND YEAR B. PHARM. DEGREE EXAMINATION - MAY 2016

SUBJECT: PHARMACEUTICAL MICROBIOLOGY (PBT 202) (CREDIT BASED SYSTEM)

Saturday, May 07, 2016

Time: 10:00 - 13:00 Hrs.

Max. Marks: 50

- Answer ALL the questions.
- Z Put question numbers properly with margin.

∠ Long Essay:

- 1. Draw neat labeled diagram of a typical bacterial cell. Discuss in detail the cell wall.
- 2. Discuss the mechanism of action, factors influencing the efficiency and applications of sterilization by ethylene oxide.
- 3. Define and classify Immunity. Explain Western blot technique with its specific application.

 $(8 \text{ marks} \times 3 = 24 \text{ marks})$

- 4A. With the help of diagrams, write a detailed note on asexual spores produced by fungi.
- 4B. Enlist the various factors affecting the course of disinfection process and explain the effect of presence of organic matter.
- 4C. Write a short note on human microbial flora.
- 4D. Write the causative agent, mode of transmission, important symptoms, prevention and treatment of malaria.

 $(4 \text{ marks} \times 4 = 16 \text{ marks})$

- 5A. Electron microscope has greater resolving power than that of optical microscope. Why?
- 5B. Define the terms autotroph, heterotroph, prototroph and auxotroph.
- 5C. Culture media containing thermolabile ingredients like gelatin can be sterilized by tyndallisation but not the injections containing thermolabile medicaments. Why?
- 5D. Write the mode of action of halogens and give one example for a halogen preparation used as antiseptics.
- 5E. Differentiate between *Escherichia coli* and *Enterobacter aerogenes* on the basis of IMViC reactions.



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SECOND YEAR B. PHARM. DEGREE EXAMINATION - MAY 2016

SUBJECT: PHARMACEUTICS (PCE 203T) (2014 REGULATION)

Tuesday, May 10, 2016

Time: 10:00 - 13:00 Hrs.

Max. Marks: 70

Answer ALL the questions.

∠ Long answer questions:

- 1. Classify Pharmaceutical powders. What are the uses of these powders? Give examples.
- 2. Explain the principle, construction and working of Sigma blade mixer with the help of diagram.
- 3. Explain controlled flocculation. Enlist any FIVE differences between Flocculated and Deflocculated suspension.

 $(10 \text{ marks} \times 3 = 30 \text{ marks})$

4. Short answer questions:

- 4A. Write the mechanism of flow of heat and its applications in Pharmacy.
- 4B. Explain the Mechanical Behavior of Solids during size reduction.
- 4C. Explain the working of a Fluid bed dryer.
- 4D. Write the advantages, disadvantages and applications of suppositories.
- 4E. Explain construction of Falling film evaporator with diagram.
- 4F. Discuss Reynolds experiment. List out various application in pharma industry.

 $(5 \text{ marks} \times 6 = 30 \text{ marks})$

5. Give reasons for the following:

- 5A. All extemporaneous dusting powders should be passed through a 100-200 mesh sieve.
- 5B. Absorbent cotton wool is treated with alkali during manufacturing.
- 5C. Calamine lotion is pink in colour.
- 5D. Size reduced particles to be used for decoction preparation.
- 5E. Repeated administration of some drugs reduces the therapeutic efficacy.

 $(2 \text{ marks} \times 5 = 10 \text{ marks})$

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SECOND YEAR B. PHARM. DEGREE EXAMINATION - MAY 2016

SUBJECT: PHARMACEUTICAL TECHNOLOGY (PCE 203) (CREDIT BASED SYSTEM)

Tuesday, May 10, 2016

Time: 10:00 - 13:00 Hrs.

Max. Marks: 50

- **Answer ALL** the questions.
- ∠ Long Essays:
- 1. Explain various methods of sterilization of catgut.
- 2. What is incompatibility? Classify incompatibilities with suitable examples. Write a note on therapeutic incompatibility.
- 3. Define fourier's law. Explain the conduction of heat through a flat slab with suitable equations.

 $(8 \text{ marks} \times 3 = 24 \text{ marks})$

Short notes:

- 4A. Classify powder. Explain tooth powders
- 4B. What is prescription? Explain the different parts of a prescription
- 4C. Describe various types of suspensions with examples
- 4D. Explain vacuum crystallizer

 $(4 \text{ marks} \times 4 = 16 \text{ marks})$

- 5A. Define throat paints. How do they differ from mouthwashes?
- 5B. Write the types of suppository bases with examples.
- 5C. Define azeotropic mixture with an example.
- .5D. Write the principle of ball mill.
- 5E. Classify dryers.

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SECOND YEAR B. PHARM. DEGREE EXAMINATION – MAY 2016

SUBJECT: PHARMACEUTICAL CHEMISTRY (PCH 204T) (2014 REGULATION)

Thursday, May 12, 2016

Time: 10:00 - 13:00 Hrs.

Max. Marks: 70

Answer ALL the questions.

∠ Long answer questions:

- 1A. Explain the chemistry of ephedrine.
- 1B. What are carotenoids? Discuss the chemistry of β -carotene.

(5+(1+4) = 10 marks)

- 2A. How do you convert aldohexose to ketohexose? Enlist the limitations of open chain structure of glucose. Write the cyclic structure of glucose and represent Alpha and Beta Glucose.
- 2B. How do you synthesize alpha-amino acids. Write any two methods.

(6+4 = 10 marks)

- 3A. Explain any one method of synthesis of pyrrole and oxazole.
- 3B. Explain nucleophilic substitution reactions of pyridine with suitable examples.
- 3C. Give the structure and uses of pyrazinamide and metronidazole.

(4+4+2 = 10 marks)

4. Short answer questions:

- 4A. Discuss the stereochemistry of di-substituted cyclohexane
- 4B. Discuss the important reactions of pyrrole and indole
- 4C. Discuss the chemistry of alpha terpineol
- 4D. Write the chemical formula and preparation of any two iron supplements used as hematinic
- 4E. What are antacids? Write the preparation of any one magnesium containing and one calcium containing antacid.
- 4F. Explain the chemistry of cardiac glycosides

 $(5 \text{ marks} \times 6 = 30 \text{ marks})$

5. Give reasons for the following:

- 5A. Camphor forms a benzylidene derivative on reaction with benzaldehyde. Justify with reaction
- 5B. In 2-bromocyclohexenone, bromine takes up the axial position rather than the equatorial position.
- 5C. Pyrimidine show less reactivity towards electrophilic aromatic reactions.
- 5D. Cholesterol is referred as C₂₇ sterol. Justify.
- 5E. Cis fatty acids have lesser melting point than corresponding trans fatty acids.



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SECOND YEAR B. PHARM. DEGREE EXAMINATION - MAY 2016

SUBJECT: PHARMACEUTICAL CHEMISTRY (PCH 204) (CREDIT BASED SYSTEM)

Thursday, May 12, 2016

Time: 10:00 - 13:00 Hrs.

Max. Marks: 50

Answer ALL questions.

∠ Long Essays:

- 1A. Explain the structural elucidation and synthesis of vitamin-A.
- 1B. What are cardiac glycosides? Give an account of cardenolides and bufadienolides.

(6+2 = 8 marks)

- 2A. Explain the stereochemistry of nitrogen compounds.
- 2B. Discuss the chemistry of Quinoline and Isoquinoline.

(4+4 = 8 marks)

- 3A. Explain Erlenmeyer's azlactone reaction for the synthesis of alpha amino acids.
- 3B. Write the structure and medicinal uses of caffeine.
- 3C. Giving suitable example explain the chemistry of Nucleotides.

(4+2+2 = 8 marks)

4. Short Essays:

4A. Explain briefly the chemistry of atropine.

(4 marks)

4B. Explain Fischer synthesis of Indole and give the structure of one Indole derivative.

(4 marks)

- 4C. i) Define heterocyclic compounds with suitable example.
 - ii) Discuss about the aromaticity of furan, pyrrole and thiophen.

(1+3 = 4 marks)

4D. How will you convert an aldose to its isomeric ketose? Explain with reactions.

(4 marks)

5. Short Answers:

- 5A. What are the basic structural features of lignans? Give their medicinal importance.
- 5B. Give the structure and uses of Quinine.
- 5C. Define any two analytical constants which are used to determine the purity of oils and fats.
- 5D. Write one method of preparation of imidazole.
- 5E. Write the structures and uses of furazolidone and INH.

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SECOND YEAR B. PHARM. DEGREE EXAMINATION - MAY 2016

SUBJECT: PHARMACEUTICAL ORGANIC CHEMISTRY (RGUHS SYLLABUS)

Thursday, May 12, 2016

Time: 10:00 - 13:00 Hrs.

Max. Marks: 80

∠ Long Essays: (Answer any TWO)

- 1. What are racemic mixtures? Explain the different methods used for the resolution of racemic mixtures.
- 2. Explain the various analytical constants used for the purity assessment of fixed oils.
- 3. Explain the stereochemistry of cyclohexane.

 $(10 \text{ marks} \times 2 = 20 \text{ marks})$

4. Short Essays: (Answer any EIGHT)

- 4A. Explain briefly the stereochemistry of oximes
- 4B. How the ring structure of glucose was established?
- 4C. Discuss the chemistry and uses of starch
- 4D. Give the structure and medicinal uses of the following:
 - i) piperazine
 - ii) sulphadiazine
 - iii) tolnaftate
 - iv) isoniazid
- 4E. How will you convert aldohexose to ketohexose and vice-versa?
- 4F. Discuss in detail about the determination of peptide structures.
- 4G. Explain the skraup synthesis of quinoline with mechanism
- 4H. Explain briefly nucleophilic substitution reactions of pyridine
- 4I. Discuss briefly the optical activity of biphenyls
- 4J. Compare the basicities of pyridine, pyrrole and alkylamines

 $(5 \text{ marks} \times 8 = 40 \text{ marks})$

5. Short Answers: (Answer ALL questions)

- 5A. Define the term atropisomerism.
- 5B. Give two uses of naphazoline.
- 5C. Define the term rancidity.
- 5D. Give the structure and use of indomethacin.
- 5E. Write the name and structure of two essential aminoacids.

- 5F. What are polynuclear hydrocarbons?
- 5G. Give the structure and use of diethylcarbamazine.
- 5H. Define the term walden inversion.
- 5I. Give the structures of thiazole and oxazole
- 5J. What is tschitschibarbin reaction?

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SECOND YEAR B. PHARM. DEGREE EXAMINATION - MAY 2016

SUBJECT: PHARMACEUTICAL MANAGEMENT (PMA 205T) (2014 REGULATION)

Saturday, May 14, 2016

Time: 10:00 - 13:00 Hrs.

Max. Marks: 70

Answer ALL the questions.

∠ Long answer questions:

- 1. Discuss Henry Fayol's 14 principles of management.
- 2. What are the Elements of promotional mix? Discuss all in detail.
- 3. With specimen formats explain all types of Journals and a Ledger.

 $(10 \text{ marks} \times 3 = 30 \text{ marks})$

4. Short answer questions:

- 4A. Discuss Conflict handling modes.
- 4B. Discuss SWOT analysis of Indian Pharmaceutical Industry.
- 4C. What are marketing objectives and strategies for different stages of product life cycle?
- 4D. What is law of demand? Explain demand curve and demand schedule.
- 4E. What are the different pharmaceutical plant layouts? Explain diagrammatically.
- 4F. Discuss Total Quality Management with respect to pharmaceutical industry.

 $(5 \text{ marks} \times 6 = 30 \text{ marks})$

5. Give reasons for the following:

- 5A. Decision making may not be always rational.
- 5B. Define Asset and Liability.
- 5C. What are the benefits of marketing research?
- 5D. What is 'efficiency' and 'effectiveness'?
- 5E. Differentiate between sole-proprietorship and partnership.

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SECOND YEAR B. PHARM. DEGREE EXAMINATION - MAY 2016

SUBJECT: PHARMACEUTICAL ANALYSIS (PQA 205) (CREDIT BASED SYSTEM)

Saturday, May 14, 2016

Time: 10:00 - 13:00 Hrs.

Max. Marks: 50

∠ Long essay:

- 1A. Write a short note on "indicator errors" in neutralization titration.
- 1B. In the titration of 0.1M hydrochloric acid against 0.1M sodium hydroxide, if methyl red (pT = 5.5) is used as an indicator, calculate the indicator error involved, state whether methyl red is a suitable indicator or not?

(5+3 = 8 marks)

2. Explain Nernst equation and factors affecting the same with relevant example.

(8 marks)

- 3A. Write note on inorganic precipitant with examples.
- 3B. Explain drying and ignition of precipitants.

(4+4 = 8 marks)

- 4A. Define the term calibration. Explain the correction factors to be considered in the calibration of graduated volumetric glassware.
- 4B. Explain the factor affecting the stability constant of metal-EDTA complexes.
- 4C. Classify solvents used in non-aqueous titrations with two examples each class.
- 4D. Explain the preparation and standardization of 0.1 M Silver Nitrate solution.

 $(4 \text{ marks} \times 4 = 16 \text{ marks})$

Short answer:

- 5A. Explain principle and reaction for the standardization 0.1M sodium nitrite.
- 5B. Write a note on crucibles used in gravimetry.
- 5C. What do you mean by absolute and derived standards?
- 5D. Name any two methods to calculate equivalent weights in redox titrations.
- 5E. What is Le Chatelier- Braun principle? Explain.

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SECOND YEAR B. PHARM. DEGREE EXAMINATION - MAY 2016

SUBJECT: PHARMACOGNOSY - II (PCO 206T) (2014 REGULATION)

Thursday, May 05, 2016

Time: 10:00 - 13:00 Hrs.

Max. Marks: 70

∠ Long answer questions:

- 1. Give the source, chemical constituents and uses of Ginger. Describe its morphology and microscopy with neat labelled diagram.
- 2. Give the Pharmacognostic report of Clove.
- 3. Describe in detail Droplet Counter Current Chromatography and HPTLC.

 $(10 \text{ marks} \times 3 = 30 \text{ marks})$

4. Short answer questions:

- 4A. What is Colophony? Describe the collection and preparation.
- 4B. Explain Stas Otto procedure for extraction of Glycosides. Add a note on identification tests of Glycosides
- 4C. Umbelliferous fruits
- 4D. Discuss briefly various methods of conservation of medicinal plants
- 4E. What are bitters? Discuss briefly Quassia and Kalmegh
- 4F. Give the source, chemical constituents and uses of Garlic and Crocus (Saffron)

 $(5 \text{ marks} \times 6 = 30 \text{ marks})$

5. Give reasons for the following:

- 5A. Preseasonal immunotherapy is different from co-seasonal immunotherapy
- 5B. Brass knife is used for scraping the outer cork of Cinnamon bark
- 5C. Pyrethrins are used as insecticides
- 5D. Lemon grass oil is used for the synthesis of Vitamin A
- 5E. Intradermal sensitivity test must be carried out on the volar surface of the lower or upper arm



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SECOND YEAR B. PHARM. DEGREE EXAMINATION - MAY 2016

SUBJECT: PHARMACOGNOSY - I (PCO 206) (CREDIT BASED SYSTEM)

Thursday, May 05, 2016

Time: 10:00 - 13:00 Hrs.

Max. Marks: 50

- Answer ALL the questions.
- ✓ Draw neat labeled diagrams and structures wherever necessary.
- ∠ Long Essays:
- 1A. Describe the morphology and microscopy of a rhizome drug with neat labeled diagram.
- 1B. Powder characters of fennel.

(6+2 = 8 marks)

2. Explain mutation and polyploidy in detail.

(4+4 = 8 marks)

3. Give the source, constituents and method of preparation and uses of tragacanth and Acacia.

(8 marks)

- 4. Short Essays:
- 4A. Source and uses of Gelatin and Spirulina
- 4B. Explain various leaf constants
- 4C. Classify and give the general tests for tannins
- 4D. Source and preparation of Castor oil

 $(4 \text{ marks} \times 4 = 16 \text{ marks})$

- 5. Short Answers:
- 5A. Natural Colour from insect
- 5B. General tests for Alkaloids
- 5C. Define Glycosides and give the examples
- 5D. Anamocytic stomata and anisocytic stomata
- 5E. Alphabetical classification