Question Paper

Exam Date & Time: 08-Jul-2023 (02:00 PM - 05:00 PM)



MANIPAL ACADEMY OF HIGHER EDUCATION

Physical Pharmaceutics - II [PCE-BP403T-S1]

Physical Pharmaceutics - II [PCE-BP4031-31]				
Marks: 75		Duration: 180 mins		
	I Multiple Choice Questions (MCQs)			
Answer all the	e questions. Se	ection Duration: 30 mins		
1)	The criterion to call a system "Colloid" is	(1)		
	A fine state of subdivision of dispersed phase Dispersed particles are in the size range of 1nm to 1µm Interface is very extensive The presence of dispersed phase in a dispersion medium			
2)	Addition of alcohol to a hydrophilic colloid lead to:	(1)		
	Crystallization Hydration Precipitation Stabilization			
3)	Which one of the following colloids is difficult to prepare?	(1)		
	Association Hydrophilic Hydrophobic Lyophilic			
4)	Which of the following properties of a particle significantly affects the physical, chemica biological properties of the drug?	l and (1)		
	Density Sedimentation Size Surface area			
5)	Stokes law cannot be used, if the Reynolds number is more than	(1)		
	0.2 1.8 9.0 18.0			
6)	The term 'Light' as applied to pharmaceutical powders means	(1)		
	Low bulk density Low granule density Low true density			

	Slightly coloured	
7)	Which one of the following dosage forms exhibit faster rate of reaction under normal conditions?	(1)
	Emulsions Ointments Solutions Suspensions	
8)	Usually, the rate of a chemical reaction may be enhanced by	(1)
	Cooling the reaction mixture Increasing the rate of stirring Raising the temperature of the reaction mixture Using stoichiometric quantities of each reactant	
9)	Which of the following expression is correct for the determination of shelf life for a first order decomposition	(1)
	t1/2 = 0.105/k $t90 = 0.693/k$ $t1/2 = 0.693/k$ $t90 = 0.105/k$	
10)	The critical value of zeta potential (in milli volts) for a stable colloid (except gold sol) is:	(1)
	From 20 to 50 From 50 to 100 Less than 20 More than 100	
11)	The angle of repose values are utilized to	(1)
	Measure the movement of granules from hopper to the table of tableting/capsule machine Select proper containers for capsules of a given mass of powders Study the absorption of drug Understand dissolution of medicament	
12)	Porosity of a porous powder is defined as	(1)
	Bulk Volume/Void Volume Void Volume/Bulk Volume Void Volume/ True Volume True Volume/Bulk Volume	
13)	When coulter- counter apparatus is employed for powder analysis, the following criterion is important	(1)
	Dispersion medium should be coloured Dispersion medium should be conducting Suspended particles should be charged Suspended particles should be spherical	
14)	Plug flow is NOT observed in cone and plate viscometer. The reason is	(1)

Slightly coloured

	Cleaning and filling of sample is easy Rate of shear is independent of the radius Shear can be maintained uniformly Temperature can be maintained uniformly			
15)	The falling sphere viscometer can be used over a range of 0.5 to 200000 poise 1 to 2 poise	(1)		
16)	200 to 2000 poise 20 to 200 poise Flocculated suspensions exhibit the flow of a type	(1)		
.5,	Dilatant Newtonian Plastic Pseudoplastic	(.,		
17)	A wetting agent is included in the formulation of a suspension, particularly when the suspended particles:	(1)		
	Are hydrophobic Are more denser than the vehicle Are water soluble Have lesser interfacial tension			
18)	For a flocculated suspension, one of the following criteria is satisfied or established.	(1)		
	High inter - particle repulsions Strong inter - particle attractions Weak inter - particle attractions Weak inter - particle repulsions			
19)	For a stable emulsion, the phase volume ratio is generally about	(1)		
	26/74 52/48 74/26 74/100			
20)	Which one of these methods is the MOST effective in preventing the rate of hydrolysis?	(1)		
	Buffer Complexation Removal of water Suppression of solubility			
II Long Answers Answer all the questions.				
Answer all the	questions.			
1)	What are various methods to determine particle size? Explain any one in detail.	(10)		
2)	What is meant by controlled flocculation? Discuss the various means by which controlled flocculation can be achieved.	(10)		
	III Short Answers			

Answer all the questions.

1)	Classify and discuss the characteristics of different colloids.	(5)
2)	Explain the principle of cup and bob viscometer with a labelled diagram	(5)
3)	Define Thixotropy. Explain its principle with suitable examples.	(5)
4)	Explain the causes for the instability of emulsion.	(5)
5)	Explain zero-order and first-order rate reactions with suitable examples.	(5)
6)	Define Micromeritics. Explain various applications of Micromeritics.	(5)
7)	Explain the effect of hydrolysis on the stability of drugs with their preventive methods.	(5)

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