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Manipal Institute of Technology, Manipal

(A Constituent Institute of Manipal University)



IV SEMESTER B.TECH (OPEN ELECTIVE - 1)
END SEMESTER EXAMINATIONS, JUNE/JULY 2016
SUBJECT: CORROSION ENGINEERING [MME 3281]
REVISED CREDIT SYSTEM

Time: 3 Hours

MAX. MARKS: 50

Instructions to Candidates:

- ❖ Answer **ALL** the questions.
- ❖ Missing data may be suitably assumed.

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| 1A. Discuss the use of vapour phase inhibitors, scavengers and hydrogen evolution poisons in controlling corrosion. | 3 |
| 1B. Explain in brief any six objectives of coatings in corrosion control. | 3 |
| 1C. Explain any four reasons why corrosion study is important? | 4 |
| 2A. How is a galvanic cell different from an electrolytic cell? | 3 |
| 2B. Differentiate activation polarization from concentration polarization. | 3 |
| 2C. Illustrate and explain the mechanism of fretting corrosion. List any eight preventive techniques. | 4 |
| 3A. State why corrosion rate measurement is necessary? What are the requirements for a good corrosion rate expression? | 3 |
| 3B. What are the guidelines to be followed while recording data during corrosion tests? | 3 |
| 3C. Explain Uniform corrosion, its mechanism and any four techniques that can be used to prevent it. | 4 |
| 4A. How anodic protection can be employed to protect a structure from corrosion? | 3 |
| 4B. List any six design rules that can be followed to prevent corrosion. | 3 |
| 4C. Discuss the construction and working of copper – copper sulfate electrode and silver-silver chloride electrode. List the requirements for a reference electrode | 4 |

- 5A.** Explain the mechanisms of Pitting Corrosion. **3**
- 5B.** How polarization curves can be used to determine the rate of corrosion? **3**
Explain the setup with the help of a neat sketch.
- 5C.** Draw a Pourbaix diagram for Aluminium in an aqueous system. Explain the different zones in it. List the uses and limitations of such diagrams. **4**