

Reg. No.



# Manipal Institute of Technology, Manipal

(A Constituent Institute of Manipal University)



## IV SEMESTER B.TECH (OPEN ELECTIVE - 1)

END SEMESTER EXAMINATIONS, JUNE/JULY 2016

SUBJECT: CORROSION ENGINEERING [MME 3281]

REVISED CREDIT SYSTEM

Time: 3 Hours

MAX. MARKS: 50

### Instructions to Candidates:

- ❖ Answer **ALL** the questions.
- ❖ Missing data may be suitably assumed.

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|-----|---|---|
| 1A. | Discuss the use of vapour phase inhibitors, scavengers and hydrogen evolution poisons in controlling corrosion.   | 3 |
| 1B. | Explain in brief any six objectives of coatings in corrosion control.   | 3 |
| 1C. | Explain any four reasons why corrosion study is important?  | 4 |
| 2A. | How is a galvanic cell different from an electrolytic cell?   | 3 |
| 2B. | Differentiate activation polarization from concentration polarization.  | 3 |
| 2C. | Illustrate and explain the mechanism of fretting corrosion. List any eight preventive techniques.   | 4 |
| 3A. | State why corrosion rate measurement is necessary? What are the requirements for a good corrosion rate expression?  | 3 |
| 3B. | What are the guidelines to be followed while recording data during corrosion tests?   | 3 |
| 3C. | Explain Uniform corrosion, its mechanism and any four techniques that can be used to prevent it.  | 4 |
| 4A. | How anodic protection can be employed to protect a structure from corrosion?  | 3 |
| 4B. | List any six design rules that can be followed to prevent corrosion.  | 3 |
| 4C. | Discuss the construction and working of copper – copper sulfate electrode and silver-silver chloride electrode. List the requirements for a reference electrode | 4 |

- 5A.** Explain the mechanisms of Pitting Corrosion. **3**
- 5B.** How polarization curves can be used to determine the rate of corrosion? **3**  
Explain the setup with the help of a neat sketch.
- 5C.** Draw a Pourbaix diagram for Aluminium in an aqueous system. Explain the different zones in it. List the uses and limitations of such diagrams. **4**