



(Manipal University)

SUBJECT: ORGANIC CHEMISTRY (CH 231)

(BRANCH: BIOMEDICAL)

Friday, 25 November 2016

Max. Marks: 100

✓ Answer ANY FIVE full questions.

- 1A.i) What is hybridization? Explain the structure of benzene based on this concept.
 ii) Explain the formation of sigma and pi-bonds with suitable examples.
- B. Give the orbital structure of the following compounds.
 i) Ethane
 ii) Methyl chloride
 iii) Acetone
- C. What are the characteristics of Vander Waals force? Explain the types of Vander Waals forces with suitable examples.

(8+6+6)

- i) Homolytic fission ii) Carbanions iv) Electrophile

(8+6+6)

- C. Give an account of different types of Elimination reactions with suitable examples.

(8+6+6)

- ii) Dehydrogenation of alcohols.

(8+6+6)

- C. Give any three methods of preparation of cycloalkanes.

(8+6+6)

- 6A. i) How are Amines classified? Give example.
 ii) Explain the following reactions of amines
 a) Carbylamine reaction b) Reaction with nitrous acid c) Acylation
- B. How are the following compounds prepared from acetaldehyde?
 i) Acetic acid ii) Chloral
 iii) Ethyl alcohol
- C. Explain the concept of resonance and the rules for stability of resonating structures
 (8+6+6)
- 7A. Give the mechanism of the following reactions of Benzene.
 i) Halogenation ii) Nitration iii) Friedel-Crafts alkylation
- B. Write an account of Refining of Petroleum.
- C. Explain the following:
 i) Octane number
 ii) Huckel rule
 (8+6+6)
- 8A. Give the reduction reactions of the following:
 i) Nitromethane ii) Methyl cyanide
 iii) Acetyl chloride iv) Ethyl alcohol
- B. Give an account of isomerism in alkanes and alkenes
- C. How are the following compounds prepared from Acetylene?
 i) Oxalic acid ii) Glyoxal iii) benzene
 (8+6+6)

