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Manipal Institute of Technology, Manipal



(A Constituent Institute of Manipal University)

III SEMESTER B.TECH END SEMESTER EXAMINATIONS, NOV 2015-DEC 2016

SUBJECT: ORGANIC CHEMISTRY [CHM 201]

Time: 3 Hours

MAX. MARKS: 50

Instructions to Candidates:

- Answer any five full questions.
- Missing data may be suitable assumed.

1A.	Explain the effects of the following on the acid strength: i. Hydrogen bond. ii. Steric hinderance iii. O-effect	3	
1B.	Discuss the structure, stability and two reactions of carbocations.	3	
1C.	Differentiate the following: i. Structural and stereoisomers ii. Position and functional isomers iii. Electromeric effect and mesomeric effect iv. Enantiomers and diastereomers	4	
2A.	Explain the following terms: Dipole moment ii. Carbenes iii. Semisynthetic flavours	3	
2B.	1 1/2 in the following and exhiam volumes.		
	a. H_3C Br $AlBr_3$?		
	b. NH ₂ CH ₃ Conc. HNO ₃ Conc. H ₂ SO ₄ ?		
	c. $\frac{Br_2}{FeBr_3}$?		

3A. i. Explain the mechanism involved in the generation of nitronium ion from the conc. Sulfuric acid and conc. Nitric acid. ii. Describe the mechanism of friedel crafts alkylation. Describe the mechanism of suphonation of benzene in detail. 3 3C. i. Explain two methods of synthesis of pyrrole. ii. Explain the shape and hybridization of NH₃. **4A.** Justify the following statements: 3 i. Humans cannot use cellulose as food. ii. Quinoline is a slightly weaker base than pyridine. iii. Trans isomers are more stable than cis isomers. 4B. 3 Assign the stereocenter in the molecule as either R or S by stepwise scheme. 4C. i.Describe the formation of H₂ molecule using orbital sketch and write MO energy level diagram. ii. Explain the significance and classification of heterocyclic compounds. 5A. What are enzymes? Give the Mechanism of enzymatic reaction. 3 How are α -amino acid prepared? Describe any two methods and two important 5B. 3 reactions of α- amino acid? **5C.** Explain the following terms with examples. i. Isoelectric point ii. Mutarotation iii. Epimerization iv. Denaturation 6A. 3 Discuss the structure of sucrose. Why is it not a reducing sugar? 6B. 3 Describe the conversion of aldohexose into aldopentose. What happens when glucose treated with the following reagents? 6C. i. Br_2/H_2O ii. Con. HNO3 iii. HI/P iv. Phenyl hydrazine