

(Add Scheme)

Reg. No.

प्रज्ञानं ब्रह्म



INSPIRED BY LIFE

Manipal Institute of Technology, Manipal

(A Constituent Institute of Manipal University)



III SEMESTER B.TECH END SEMESTER EXAMINATIONS, NOV/DEC 2015

SUBJECT: ORGANIC CHEMISTRY [CHM 201]

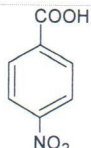
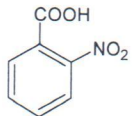
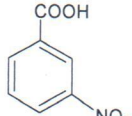
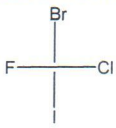
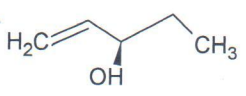
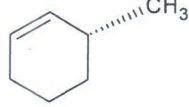
Time: 3 Hours

MAX. MARKS: 50

Instructions to Candidates:

- ❖ Answer **any five full** questions.
- ❖ Missing data may be suitable assumed.

1A.	Explain three reactions of amino and carboxyl groups in the amino acids.	3
1B.	Explain the following terms: i. Mutarotation ii. Mesomeric effect iii. Chain isomerism	3
1C.	Explain the primary, secondary, tertiary and quaternary structure of protein.	4
2A.	Discuss the structure and stability of free radicals.	3
2B.	Predict the product/s in the following and explain your reasoning. <div style="display: flex; flex-direction: column; align-items: center;"> <div style="display: flex; align-items: center; margin-bottom: 20px;"> i. <div style="text-align: center;"> <chem>CC(=O)Nc1ccc(C)cc1</chem> $\xrightarrow[\text{FeBr}_2]{\text{Br}_2}$? </div> </div> <div style="display: flex; align-items: center;"> ii. <div style="text-align: center;"> <chem>CNc1ccccc1C</chem> $\xrightarrow[\text{Conc. H}_2\text{SO}_4]{\text{Conc. HNO}_3}$? </div> </div> </div>	3
2C.	Discuss two physical and two chemical properties of starch.	4
3A.	i. Describe in mechanism of chlorination of benzene. ii. Explain Ruff's degradation with a suitable example.	3
3B.	Compare and explain the acidities of the following molecules:	3

	<div style="display: flex; justify-content: space-around; align-items: center;"> <div style="text-align: center;">  <p>i.</p> </div> <div style="text-align: center;">  <p>ii.</p> </div> <div style="text-align: center;">  <p>iii.</p> </div> </div>	
3C.	<p>i. Explain metamerism and tautomerism using suitable examples.</p> <p>ii. What are heterocyclic compounds? Explain their classification based on electronic structures with examples.</p>	4
4A.	<p>Justify the following statements:</p> <p>i. Toluene is more reactive than benzene</p> <p>ii. Amylose has linear chain structure while amylopectin has a branched one.</p> <p>iii. Pyridine is less basic than ammonia.</p>	3
4B.	<p>Assign the stereocenter in the molecule as either R or S by stepwise scheme.</p> <div style="display: flex; justify-content: space-around; align-items: center;"> <div style="text-align: center;"> <p>i.</p>  </div> <div style="text-align: center;"> <p>ii.</p>  </div> <div style="text-align: center;"> <p>iii.</p>  </div> </div>	3
4C.	<p>i. Explain D and L designations of sugar.</p> <p>ii. Describe epimerization process using suitable example.</p>	4
5A.	<p>Explain the synthesis of the following:</p> <p>i. Cimetidine</p> <p>ii. Codeine</p> <p>iii. Quinoline</p>	3
5B.	Describe three theories of enzymatic actions.	3
5C.	<p>i. Explain in detail the structure and properties of the following:</p> <p>a. Zwitterion b. Carbenes.</p> <p>ii. Explain the steps involved in the conversion of D-fructose to D-glucose.</p>	4
6A.	<p>Differentiate between the following;</p> <p>i. Enantiomers and diastereomers</p> <p>ii. Fibrous and globular proteins</p> <p>iii. Carbocations and carbanions</p>	3
6B.	Define specific rotation. Explain the working of a polarimeter using a schematic diagram.	3
6C.	<p>i. Explain the shape and hybridization of H₂O.</p> <p>ii. What is barbitol? Describe its synthesis.</p>	4