

III SEM. B.Tech. MECHATONICS ENGG. DEGREE END
SEMESTER MAKE UP EXAMINATION DEC 2015/JAN 2016

SUBJECT: MATERIALS SCIENCE AND ENGINEERING (MTE 2101)

Time: 3 Hours

MAX. MARKS: 50

Instructions to Candidates:

- ❖ Answer **ALL** the questions.
- ❖ Missing data, if any, may be assumed suitably.
- ❖ Draw neat labeled diagram wherever necessary.

- 1A. Two metal A and B have 100% mutual solubility in the liquid and solid states. The melting point of pure metals A and B are 900°C and 750°C. Details of the start of solidification and end of solidification of various alloys in the series are as follows: (05)

| Alloy of Composition | Temp. at start of solidification (°C) | Temp. at the end of solidification (°C) |
|----------------------|---------------------------------------|---|
| 90% A – 10% B | 890 | 840 |
| 70% A – 30% B | 875 | 825 |
| 50% A – 50% B | 860 | 810 |
| 30% A – 70% B | 845 | 795 |
| 10% A – 90% B | 830 | 780 |

- i. Draw the phase diagrams of the series if there are no solid state reactions and label all regions.
- ii. Predict the number, type, relative amounts and concentration of phases present in an alloy of 60% A and 40% B at 830°C and 20°C.

- 1B. Describe the addition and condensation polymerization reaction. Also, explain the termination process of both the reactions. (03)

- 1C. Discuss the stress-strain behavior for a fiber reinforced composite when the load is applied in longitudinal direction with respect to the fiber alignment. (02)

- 2A. Draw the Iron-Iron Carbide Equilibrium diagram and explain the reaction which takes place at 1148°C. (05)

- 2B. State Luminescence. Describe it in the context of LED and LASERS. (05)

- 3A. Draw the FCC structure and show the packing structure of FCC is 0.74. (03)

- 3B. Discuss the following with appropriate sketch: i. Annealing. ii. Flame Hardening. (04)

- 3C. Discuss the following: i. Frenkel Defect. ii. Schottky Defect. (03)

- 4A. Discuss the following: i. Medium Carbon Steel. ii. Gray Cast Iron. (02)

- 4B. Describe the dispersion strengthened composites and its influencing factors. (03)

- 4C.** What are Miller Indices? Represent the following Miller Indices in a simple unit cubic cell: **(05)**
- i. $[123]$
 - ii. (121)
 - iii. $[222]$
 - iv. (201)
- 5A.** Draw and explain the TTT diagram for eutectoid steel and mark the phases. **(05)**
- 5B.** Discuss the process of Chemical Vapour Deposition and Sputtering with appropriate diagram. **(05)**