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MANIPAL INSTITUTE OF TECHNOLOGY
Manipal University



FIRST SEMESTER M.TECH (ME) DEGREE END SEMESTER EXAMINATION
NOV/DEC 2015
SUBJECT: VLSI PROCESS TECHNOLOGY (ECE - 523)

TIME: 3 HOURS

MAX. MARKS: 50

Instructions to candidates

- Answer **ANY FIVE** full questions.
- Missing data may be suitably assumed.

- 1A. A constant- source boron diffusion is performed into a n-type silicon wafer at 1050 °C for an hour. Assume that surface concentration is limited by solid solubility limit. Given that surface solubility is $2.4 \times 10^{20} / \text{cm}^3$. If n-type background doping of silicon is $5 \times 10^{16} / \text{cm}^3$ find the junction depth. Given that $D_o = 1.0 \times \text{Cm}^2/\text{sec.}$, and activation energy is 3.5 eV.
- 1B. Predict the concentration of oxygen in the crystal at a fraction solidified of 0.4. Given the segregation co-efficient of oxygen is 0.25. Given that the fraction of oxygen atoms solidified at the top is $1.3 \times 10^{18} / \text{Cm}^3$. And comment on the result
- 1C. At what temperature the Ga and As vacancy densities become equal in GaAs. Given that activation energy of Ga vacancy is 0.4 eV and that for As is 0.7 eV.
- (5+2+3)
- 2A. A 1000-Å gate oxide is required for some technology. It has been decided that the oxidation will be carried-out in wet oxygen. If there is no initial oxide, for how long the oxidation be done? If the oxidation be in linear or parabolic regime or between two? Given that: $A = 0.50 \mu\text{m}$ and $B = 0.203 \mu\text{m}^2 / \text{hour}$
- 2B. Explain significance of the parameter τ in the contest of ultra-thin oxides.
- 2C. Calculate the diffusivity of arsenic in Silicon at 1000 °C if the concentration of Arsenic is much less than intrinsic carrier concentration and again if the arsenic concentration is $1 \times 10^{19} \text{cm}^{-3}$. Intrinsic carrier concentration for Si at 1000 °C is 1×10^{19} . Given that: $D_o = 0.066$ and $E_a = 3.44$, D_o^- (diffusivity for negatively ionised vacancy) = 12.0 and corresponding $E_a = 4.05$ eV.
- (4+3+3)
- 3A. Phosphorus is added to a silicon wafer from a gaseous source at 975 °C for 30 minutes. Determine the junction depth for i) $0.3 \Omega \text{ Cm}$ p-type substrate ($N_A = 6.3 \times 10^{16} \text{cm}^{-3}$) and ii) $20 \Omega\text{-cm}$ p-type substrate ($N_A = 6.8 \times 10^{18} \text{Cm}^{-3}$). Diffusion co-efficient of Phosphorous is $10^{-13} \text{cm}^2 \text{S}^{-1}$ and that its solid solubility is 10^{21}cm^{-3} at 975 °C.
- 3B. Write short note on: i) Extrinsic -diffusion of dopant atoms. ii) Emitter push effect in silicon n-p-n transistors.
- (5+5)
- 4A. Compare and contrast optics of contact, proximity and projection printers and areal image formation
- 4B. A proximity printer operates with a $10 \mu\text{m}$ mask-wafer gap and a wavelength of 430 nm. Another printer uses $40 \mu\text{m}$ gap and wavelength 250nm. Which offers higher resolution?

4C. . Why must the mask used in a wafer stepper be completely be free from defects while some defects be tolerated in systems exposing entire wafer at once?

(5+3+2)

5A. State the chemical reactions that are relevant for the Silicon growth by chemical vapour deposition technique?

5B. Explain different methods of wafer cleaning and native oxide removal before initiating thin film growth process.

5C. Explain limitations of CV technique for impurity profile determinations

(3+3+4)

6 Answer the following with proper explanation.

i) Low-pressure CVD is ---- controlled reaction whereas atmospheric CVD is ----- controlled reaction.

ii) If a process relies on ion-bombardment of the wafers, would you want to put the wafers on the electrode with --- surface area electrode in the plasma-chamber.

iii) How CVD techniques score over PVD techniques?

iv) State two accurate optical techniques to measure the thickness of oxide.

v) Explain a method to purify hydrogen to be used in pre-clean methods of VPE.

(2*5)

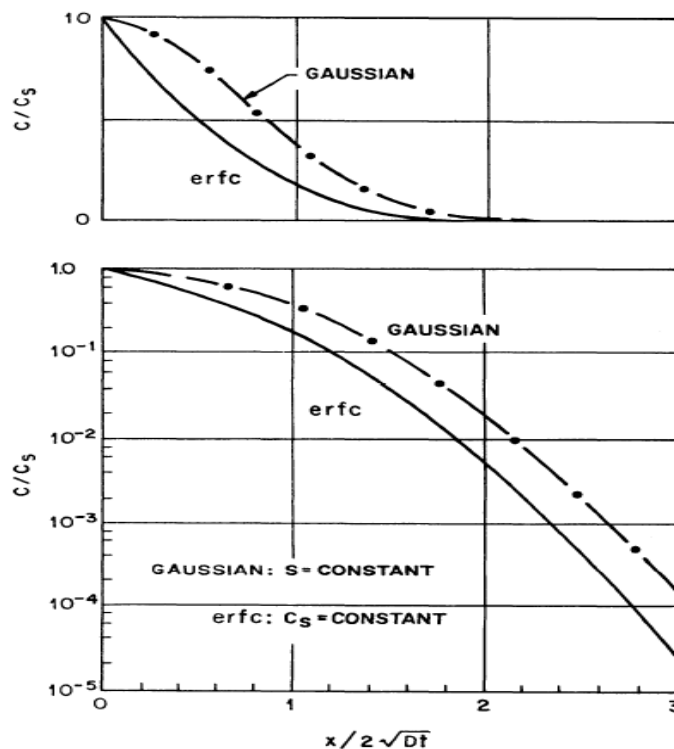


Figure : Normalised concentration verses normalised distance for erfc and Gaussian functions.