

VII SEMESTER B.TECH (CHEMICAL ENGG.)
END SEMESTER EXAMINATIONS, NOV/DEC 2015
SUBJECT: PROCESS MODELLING AND SIMULATION (CHE -403)
REVISED CREDIT SYSTEM

Time: 3 Hours

MAX. MARKS: 100

Instructions to Candidates:

- ❖ Answer **ANY FIVE FULL** the questions.
- ❖ Missing data may be suitably assumed.
- ❖ Draw IFD wherever necessary.

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| 1A. | The fluid is flowing along the inclined flat plate which is at angle β from the vertical. The film thickness over the plate is δ and the fluid is moving with velocity V . Develop an equation for velocity distribution. State all the assumptions. | 10 |
| 1B. | An agitated 2m^3 tank is initially filled with water at 25°C . A steam coil with an area of 0.5 m^2 heats the water using steam condensing at 120°C . The heat transfer coefficient for heating is $1140\text{ J/s.m}^2.\text{K}$. The tank losses heat to the surrounding through the walls which have an area of 9 m^2 . The heat transfer coefficient for heat losses is $10\text{ J/s.m}^2.\text{K}$ and the surrounding temperature is 25°C . How long will it take to heat the water to boiling point. Neglect the heat capacity of the tank walls. (Take water $C_p=4187\text{J/Kg.K}$) | 10 |
| 2A. | Derive the center difference technique and develop the mathematical model or the dynamics response of an unsteady state counter current plug flow heat exchanger. Give briefly the solution procedure. | 15 |
| 2B. | List out the characteristics of model.(Any 5 only) | 05 |
| 3A. | Develop steady state tray composition for a 6 plate absorption column. A linear equilibrium relation holds between liquid x_m and vapor y_m on each plate and is given by $y_m = ax_m + b$. The inlet compositions to the column along with liquid and gas flow rate are known. Briefly give the solution procedure. | 07 |
| 3B. | Write a short notes on degree of freedom? | 05 |
| 3C. | <p>Consider a set of reversible reaction:</p> $A \xrightleftharpoons[K_2]{K_1} B \text{ and } B \xrightleftharpoons[K_4]{K_3} C$ <p>Where N_A, N_B, N_C be the moles of A, B, C respectively present at any time t. Since the reaction is constant volume, N_A, N_B, N_C are proportional to concentration. Derive</p> $\frac{d^2 N_A}{dt^2} + (K_1 + K_2 + K_3 + K_4) \frac{dN_A}{dt} + ((K_1 * K_3) + (K_2 * K_4) + (K_1 * K_4))N_A - (K_2 * K_4) = 0$ | 08 |

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| 4A. | Write the step by step procedure for modeling of any process. | 10 |
| 4B. | Define the following (i) Model (ii) simulation (iii) grey box model (iv) white box model (v) black box model | 5 |
| 4C. | Differentiate between lumped and Distributed parameter model with examples | 5 |
| 5 | Write the general modeling equations for a ternary equilibrium column of five stages including condenser and reboiler for a multi-component distillation column. Draw a neat information flow diagram for feed tray , reboiler and Condensor. | 20 |
| 6 | Develop the design equations for multi component pipe line flasher. Discuss the model for establishing temperature. | 20 |
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