



# MANIPAL INSTITUTE OF TECHNOLOGY

MANIPAL

*A Constituent Institution of Manipal University*

## FIFTH SEMESTER B.TECH. (INSTRUMENTATION AND CONTROL)

END SEMESTER EXAMINATIONS, DEC 2016/JAN 2017

SUBJECT: ANALYTICAL INSTRUMENTATION [ICE-321]

Time: 3 Hours

MAX. MARKS: 50

### Instructions to Candidates:

- ❖ Answer **ANY FIVE FULL** questions.
- ❖ Missing data may be suitably assumed.

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|------------|---|----------|
| <b>1A.</b> | Derive Beer-Lambert law and explain chemical deviations of Beer- lambert law.   | <b>5</b> |
| <b>1B.</b> | Explain different types of monochromators which are used in spectrophotometers.   | <b>3</b> |
| <b>1C.</b> | The molar absorptivity of the iron (II)-2,2',2"-terpyridyl complex is $1.11 \times 10^4$ at 522 nm. Calculate the concentration of the complex in a solution which has a percent transmittance of 36.5 at 522 nm in a cell with a path length of 1.00 cm. | <b>2</b> |
| <b>2A.</b> | Draw optical diagram for double beam UV and Visible spectrophotometry. Explain photomultiplier tube.  | <b>4</b> |
| <b>2B.</b> | Explain the infrared spectrophotometer based on Attenuated Total Reflectance technique.   | <b>4</b> |
| <b>2C.</b> | What is the role of atomizer in Flame photometers?  | <b>2</b> |
| <b>3A.</b> | What is Chromatography? Briefly explain the working of High Pressure Liquid Chromatograph.  | <b>5</b> |
| <b>3B.</b> | With neat diagram, explain atomic absorption spectrophotometer.   | <b>3</b> |
| <b>3C.</b> | Explain principle of time of flight mass spectrometer.  | <b>2</b> |
| <b>4A.</b> | Explain the technique used for the measurement of average spacing between rows of atoms.  | <b>4</b> |
| <b>4B.</b> | Explain ion selective electrode which is used to measure silicon in a solution  | <b>4</b> |
| <b>4C.</b> | With a neat diagram explain digital type pH meter.  | <b>3</b> |
| <b>5A.</b> | Explain the rules for finding the net spin of a nucleus and with neat diagram explain continuous wave NMR Spectroscopy.   | <b>5</b> |
| <b>5B.</b> | Name the method of oxygen measurement based on curie's law and explain its working.   | <b>5</b> |
| <b>6A.</b> | With neat diagram, explain measurement of nitrogen using thermal conductivity analyzer.   | <b>4</b> |
| <b>6B.</b> | Explain the measurement of Carbon monoxide in a gas using infrared radiations   | <b>3</b> |
| <b>6C.</b> | Explain the detection of ' $\alpha$ ' particle using any one of the crystal scintillator.   | <b>3</b> |

