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DEPARTMENT OF SCIENCES, M. Sc. (P/C/M/G) II SEMESTER END SEMESTER EXAMINATIONS JUNE 2017 Subject: Quantum Mechanics II (PHY-606) (REVISED CREDIT SYSTEM)

Time: 3 Hours Date: June 2017 MAX. MARKS: 50

Note: (i) Answer any five full questions.

(ii) Answer the questions to the point.

1. (i) If the eigenvalues of J^2 and J_z are given by $J^2|\lambda, m >= \lambda|\lambda, m >$ and $J_z|\lambda, m >= m|\lambda, m >$. For the simultaneous eigenvector of J^2 and J_z , $|\lambda, m >$ show $\lambda \ge m^2$. Take $\hbar = 1$. [3]

(ii) Show that $\vec{L} \times \vec{L} = i\vec{L}$. [4]

(iii) Determine the orbital momenta of two electrons in the configuration p^1d^1 . [3]

2. (i) Calculate the energy correction due to spin-orbit coupling for hydrogen atom. [5]

(ii) Given the matrix for H^0 and H' as $\begin{pmatrix} E_0 & 0 \\ 0 & E_0 \end{pmatrix} \begin{pmatrix} 0 & -A \\ -A & 0 \end{pmatrix}$ in the orthonormal basis |1 > and |2 >, determine the

and (b) energy eigenfunctions. [3]

3. (i) The result of the variational method always gives an upper limit for the ground state energy of the system. Why? [4]

(ii) Use the WKB method to calculate the transmission coefficient for the potential barrier

$$V(x) = \begin{cases} V_0 - ax, & x > 0; \\ 0, & x < 0. \end{cases}$$
[6]

4. (i) Use time dependent perturbation theory to obtain an expression for the transition in first order approximation. [5]

(ii) Calculate the electric dipole transition moment $\langle 2p_z | \mu_z | 2s \rangle$ for the $2s \rightarrow 2p_z$ transition in a hydrogen atom. [5]

5. (i) Obtain the expression of differential scattering cross-section in terms of beam luminosity. [4] (ii) In scattering from a potential V(r); the wave function $\psi(r)$ is written as an incident plane wave plus an outgoing scattered wave: $\psi = exp(ikz) + f(r)$. Derive a differential equation for f(r) in the first Born approximation. [6]

6. (i) Explain how the KG equation leads to positive and negative probability density values? [5]

(ii) Obtain the plane wave solutions of the Dirac equation.

$$\lfloor 5 \rfloor$$

Useful formulae:

$$\nabla^{2}t = \frac{1}{r^{2}}\frac{\partial}{\partial r}\left(r^{2}\frac{\partial t}{\partial r}\right) + \frac{1}{r^{2}sin\theta}\frac{\partial}{\partial \theta}\left(sin\theta\frac{\partial t}{\partial \theta}\right) + \frac{1}{r^{2}sin^{2}\theta}\frac{\partial^{2}t}{\partial \phi^{2}}$$
$$\int_{0}^{\infty}exp(-a^{2}x^{2})cos(bx)\,dx = \frac{\sqrt{\pi}}{2a}exp\left(-\frac{b^{2}}{4a^{2}}\right)$$
$$\int_{0}^{\infty}x^{n}exp(-ax)\,dx = \frac{n!}{a^{n+1}}, \quad \text{where} \quad n \ge 0, \quad a > 0$$

For hydrogen atom:

$$\psi_{1s} = \left(\frac{1}{\pi a_0^3}\right)^{\frac{1}{2}} exp\left(-\frac{r}{a_0}\right)$$
$$\psi_{2s} = \left(\frac{1}{32\pi a_0^3}\right)^{\frac{1}{2}} \left(2 - \frac{r}{a_0}\right) exp\left(-\frac{r}{2a_0}\right)$$
$$\psi_{2p_z} = \left(\frac{1}{32\pi a_0^3}\right)^{\frac{1}{2}} \frac{r}{a_0} exp\left(-\frac{r}{2a_0}\right) cos\theta$$