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## INTERNATIONAL CENTRE FOR APPLIED SCIENCES

(Manipal University)

III SEMESTER B.S. DEGREE EXAMINATIONS - OCT. /NOV. 2017

SUBJECT: ORGANIC CHEMISTRY (CH 231)

(BRANCH: BIOMEDICAL)

Saturday, 4 November 2017

Time: 3 hrs.

Max. Marks: 100

✓ Answer ANY FIVE full questions.

1 A. Explain the following:

- i) Formation of a Pi bond
- ii) Hybridisation of the carbon atom in Ethylene molecule.
- iii) Intramolecular hydrogen bonding
- iv) Isomerism in Butene

B. Give an account of the following:

- i) Inductive effect
- ii) Mesomeric effect

C. Explain the fractionation of petroleum and the important applications of different fractions of petroleum.

(8+6+6)M

2A. How are the following conversions made?

- i) Isopentane to n-pentane
- ii) Isobutene to isooctane.
- iii) Propylene to acetic acid
- iv) Propene to 1-propanol

B. Give three different methods of preparation of alkyl halides from alcohols.

C. Explain the mechanism of the following:

- i) Anti Markovnikoff's addition reaction.
- ii) Dehydration of primary and tertiary alcohol's.

(8+6+6)M

3A. Give an account of the following reactions

- i) Mechanism of chlorination of methane
- ii) Differences between  $S_N1$  and  $S_N2$  mechanisms

B. Explain the following with suitable examples:

- i) Wurtz reaction
- ii) Williamson's synthesis

C. How do you prepare acetic acid from the following?

- i) Grignard reagent.
- ii) Ethyl alcohol.
- iii) Methyl bromide

(8+6+6)M

4A. Give reasons for the following:

- i) Aldehydes are more reactive than Ketones.
- ii) Aldehydes but not ketones give positive test with Tollen's reagent.
- iii) Ethyl alcohol but not methyl alcohol gives iodoform test.
- iv) Denatured spirit is not absolute alcohol.

B. Explain the manufacture of ethyl alcohol from starch.

C. Give any three methods of preparation of ethylene glycol.

(8+6+6)M

5A. Explain the following:

- i) Hydrolysis of alkyl cyanide
- ii) Hydrolysis of ester
- iii)  $\alpha$ -halogenation of carboxylic acid
- iv) Kolbe's reaction

B. Write the chemical reaction of Acetic acid with the following reagents.

- i) Thionyl chloride
- ii) Ammonia and heat
- iii) Ethyl alcohol and  $H_2SO_4$ .

C. How are the following conversions affected?

- i) 2-Butene to 1,2-Dimethylcyclopropane
- ii) Propyne to Propene
- iii) Acetylene to oxalic acid.

(8+6+6)M

6A. Illustrate the following:

- i) Wolf-Kishner reduction
- ii) Carbylamine reaction
- iii) Cannizzaro reaction
- iv) Friedel-Crafts acylation

B. Explain the following with suitable examples:

- i) Oxidation of alcohols
- ii) Dehydrogenation of alcohols

C. Write the reaction of Ethylamine with the following reagents.

- i) Methyl iodide
- ii) Nitrous acid
- iii) Hydrochloric acid

(8+6+6)M

7 A. Explain the structure of Benzene in terms of classical theory and M.O. theory.

B. Give three different ring opening reactions of cyclopropane.

C. Write an account of the following:

- i) Conditions for Aromaticity
- ii). Method of separation of mixture of amines

(8+6+6)M

8 A. Give an account of the following

- i) Refining of Petroleum
- ii) Cracking of Petroleum

B. Explain the preparation of methane from the following

- i) Sodium acetate
- ii) Methyl magnesium iodide
- iii) Methyl iodide

C. Explain the following:

- i) Saytzeff rule
- ii) Octane number

(8+6+6)M

