


VI SEMESTER B.Tech. (CHEMICAL ENGINEERING)
MAKE UP EXAMINATION, JUNE 2018
SUBJECT: O.E.: INDUSTRIAL POLLUTION CONTROL (CHE 3282)
REVISED CREDIT SYSTEM

Time: 3 Hours

MAX. MARKS: 50

Instructions to Candidates: Answer ALL the questions.

1 A	Explain the Hydrologic cycle with a neat diagram.	3
1 B	What is the significance of the following? i) Volatile solids ii) Turbidity iii) Temperature iv) Alkalinity v) Acidity vi) BOD	3
1 C	Explain the principle of i) Freeze out sampling & ii) Impingers to collect gaseous sample and iii) High volume sampler & iv) Cyclone separator to collect particulate sample.	4
2 A	Explain i) Advanced oxidation process ii) Activated Carbon Adsorption used in tertiary treatment of wastewater.	4
2 B	Differentiate between Anaerobic and Aerobic treatment of wastewater. (Any 2)	2
2 C	Explain with neat diagrams the principle and working of i) Primary Sedimentation Tank & ii) Dispersed Air Floatation in primary treatment of wastewater.	4
3 A	Describe the three approaches for capture of CO ₂ involved in Carbon sequestration.	3
3 B	Explain with a neat diagram the principle and working of i) Electrostatic Precipitator & ii) Bag filter used to control particulate matter emission.	4
3 C	The maximum CO concentrations normally measured in downtown Salt Lake City are about 3000 µg/m ³ . These values occur during strong inversions during which wind speed was 0.5 m/s and mixing height was 95 m. The background concentration for this situation is estimated to be 450 µg/m ³ . The downtown area of Salt Lake City may be approximated as 4 km by 3 km square. Estimate the emission density (g/s. m ²) for CO for downtown Salt Lake City.	3
4	Explain the different steps involved in sludge treatment (in not more than two or three sentences per process).	10
5 A	Explain any two methods of managing e-waste in industry.	4
5 B	Describe methods adopted to control noise pollution in industry.	6

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