



B.TECH. (CHEMICAL ENGINEERING)

MAKE UP EXAMINATIONS, Dec 2018

SUBJECT: COMPUTATIONAL METHODS IN CHEMICAL ENGINEERING [CHE 3105]

REVISED CREDIT SYSTEM,

Time: 3 Hours

Instructions to Candidates

MAX MARKS: 50

- ❖ Answer all questions **FULLY**.
- ❖ Missing data may be suitable assumed.

1A	Solve for X by Gauss Elimination method when AX =B <div><div>A>>></div><table><tr><td>2</td><td>3</td><td>1</td><td>4</td></tr><tr><td>1</td><td>2</td><td>3</td><td>2</td></tr><tr><td>3</td><td>1</td><td>2</td><td>3</td></tr><tr><td>3</td><td>4</td><td>5</td><td>3</td></tr></table><div>B>>></div><table><tr><td>22</td></tr><tr><td>15</td></tr><tr><td>26</td></tr><tr><td>26</td></tr></table></div>	2	3	1	4	1	2	3	2	3	1	2	3	3	4	5	3	22	15	26	26	4
2	3	1	4																			
1	2	3	2																			
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1B	f(x)= xe ^x - cos(x) =0 Solve using Newton Raphson method. Initial guess is 0.6.Make 3 iterations	3																				
1C	<div>$0 = \frac{1}{\sqrt{f}} + 2.0 \log \left(\frac{\epsilon}{3.7D} + \frac{2.51}{Re\sqrt{f}} \right)$</div> <p>Calculate the friction factor with above equation using regular Falsi method if $\epsilon/D = 0.0001$ and $Re = 10,000$. Use lower and upper limits as 0.02 and 0.05 respectively. Make 3 iterations.</p>	3																				

2A	Solve by RK4 method, the equation dy/dx = 3e ^{-x} -0.4y Initial condition is that at x=0, y= 5, Take step size as 1.5 , Find y at x= 6	6
2B	<p>The specific heat of gas as a function of temperature is given $Cp = 0.4 + \frac{18}{T+40}$ KJ/kg. The temperature of the gas is increased from T₁ = 10⁰C to T₂ = 50⁰C by the addition of heat at constant pressure. The heat added to the gas is given by the expression:</p> <div>$Q = \int_{T_1}^{T_2} Cp \, dt$</div> <p>Determine the total heat added to the gas using Simpsons 1/3 rule, make 8 intervals.</p>	4

3A	<p>The rate of an enzymatic reaction is given by the expression: The k and Km can be estimated by linear regression by defining x =1/S and y =1/r, Find the values of K_m and k. Find the value of R² of the linear fitting.</p> <div><table><tr><td>[S]</td><td>1.233</td><td>0.540</td><td>0.442</td><td>0.258</td><td>0.198</td><td>0.162</td><td>0.130</td><td>0.128</td></tr><tr><td>r</td><td>5.970</td><td>3.319</td><td>2.253</td><td>2.547</td><td>1.493</td><td>1.182</td><td>1.095</td><td>0.869</td></tr></table></div> <div>$r = \frac{k[S]}{K_m + [S]}$</div>	[S]	1.233	0.540	0.442	0.258	0.198	0.162	0.130	0.128	r	5.970	3.319	2.253	2.547	1.493	1.182	1.095	0.869	8
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r	5.970	3.319	2.253	2.547	1.493	1.182	1.095	0.869												

3B	<p>The vapor pressure of water increase with temperature. The values are tabulated below.</p> <table><tr><td>T °C</td><td>30</td><td>40</td><td>50</td><td>60</td></tr><tr><td>p_v (mm Hg)</td><td>31.8</td><td>55.3</td><td>92.5</td><td>149.4</td></tr></table> <p>Find the vapor pressure at 45 °C using Newton's forward interpolation</p>	T °C	30	40	50	60	p _v (mm Hg)	31.8	55.3	92.5	149.4	2
T °C	30	40	50	60								
p _v (mm Hg)	31.8	55.3	92.5	149.4								
4A	<p>Solve the system of simultaneous non-linear equations using Newton-Raphson method</p> <p>$x^2+y=11$ $x+y^2=7$</p> <p>Initial guess $x_0=3.5$, $y_0=-1.8$. Make 1 iterations.</p>	4										
4B	<p>Solve by RK4 method and find value of y and z at $x=0.2$</p> <p>$dy/dx=x+z$ $dz/dx=x-y^2$</p> <p>Step size $\Delta x=0.1$. Initial condition is that at $x=0$, $y=2$, $z=1$</p>	6										
5	<p>Solve the equation for flow with reaction in a PFR using finite difference implicit method</p> <div>$\frac{\partial c}{\partial t} = D \frac{\partial^2 c}{\partial x^2} - v \frac{\partial c}{\partial x} - kc$</div> <p>where C is concentration in water (mol/L), t is time (s), v is pore water flow velocity (m/s), x is distance (m), D is the hydrodynamic dispersion coefficient (m²/s), k is the first order reaction constant</p> <p>The length of reactor is 1 m. $dx=0.25$ m, $dt=1$ s. At time $t=0$, the initial concentration is 0 at all nodes except left boundary where the boundary condition is concentration is 1 mol/L. The other end of the rod has zero dispersion boundary condition.</p> <p>Pore Velocity = 0.25 m/s, $k=0.3\text{ s}^{-1}$, Dispersion coefficient $D=0.3\text{ m}^2/\text{s}$, Take $dx=0.25$ m, Find the concentrations at the nodes at time =1 seconds</p>	10										