

IV SEMESTER B.TECH. EXTERNAL EXAMINATIONS APRIL/MAY 2019

SUBJECT: HEAT AND MASS TRANSFER OPERATIONS IN

BIOPROCESSING [BIO2205]

Date of Exam: 02/05/2019 Time of Exam: 2.00 PM – 5.00 PM Max. Marks: 50

Instructions to Candidates:

✤ Answer ALL the questions & missing data may be suitable assumed

1A.	A truncated cone 0.3 m high is constructed of Aluminum. The diameter at the top is 75 mm and diameter at the bottom is 125 mm. The lower surface is maintained at 93^{0} C, the upper surface is at 540^{0} C. The outer surface is insulated. Assuming one dimensional heat flow, determine the rate of heat transfer in Watts, $K_{Al} = 230$ W/mK	4
1B.	Prove that for a spherical surface covered with an insulation of thermal conductivity k, critical insulation radius is 2k/h, where h is the convective heat transfer coefficient outside the insulation	3
1C	Water flows counter-currently through two concentric pipes with inner pipe of 1" OD and outer pipe of 2"OD. Ignore the thickness of inner pipe for calculation purpose. Cold water enters inner pipe at 1m/s at 30 ^o C and exits at 60 ^o C. Hot water enters outer pipe at 90 ^o C at a velocity of 0.5 m/s. Compute LMTD. $C_p=4180 \text{ J/kg K}, \rho = 990 \text{ kg/m}^3$	3
2A.	 Air at 3 atm and 200⁰C is heated as it flows through a tube with a diameter of 2.54 cm at a velocity of 15 m/s. i. Calculate the heat transfer per unit length of tube if the wall temperature is at 20⁰C above the air temperature all along the length of the tube and constant flux condition is maintained at the wall. ii. How much would bulk temperature increase over a 4 m length of the tube? Air properties: μ= 2.57 x10⁻⁵ kg/m.s, k = 0.0386 W/mK, C_p = 1.025kJ/kgK. 	5
2B.	Saturated steam at one atmospheric pressure condenses on a 3m height (H) and 4 m wide (W) vertical plate that is maintained at 90 ^o C by circulating cooling water through the other side. Determine, i. Total rate of heat transfer by condensation to the plate, kW ii. The average heat transfer coefficient over the entire plate kW/m ² K iii. The rate at which the condensate drips off the plate at the bottom in kg/s., Average properties of condensate : $\rho = 962 \text{ kg/m}^3$, $k = 0.677 \text{ W/mK}$, $\mu = 3x10^{-4} \text{ kg/m.s}$, latent heat = 2.27 x10 ⁶ J/kg. $h = 0.944 (\frac{k^3 \rho^2 g \lambda}{\Delta T H \mu})^{1/4}$	5

3A.	A heated vertical wall of 0.75 m high of an oven for baking food with free surface at 500K is in contact with air at 300K. Calculate heat transfer coefficient and heat transfer per meter width of wall. Note that heat transfer by radiation need not be considered. Physical properties of air $\mu = 2.32 \times 10^{-5}$ Pa.s, Pr = 0.690, k=0.065W/mK, Nu=0.59(Gr.Pr) ^{0.25}	4
	$Gr = \frac{L^{3}\rho^{2}g\beta\Delta T}{\mu^{2}}$	
3B.	Two parallel infinite plane surfaces are maintained at 200 ^o C and 300 ^o C. Determine the net rate of radiation heat transfer per unit area when (a) the two surfaces are grey with emissivity of 0.7 (b) the two surfaces are black	3
3C.	Explain different regimes of boiling and the factors affecting them –(a) Natural convection (b) Nucleate (c) Transition	3
4A.	Oxygen is diffusing through a non-diffusing layer of methane 5 mm thick. The temperature is 20° C and the pressure is 1 atm. Calculate the rate of diffusion of oxygen in kg/h through 1 m ² of methane film when the concentration of oxygen on two sides of the methane film are 15% and 5% by volume respectively. The diffusivity of oxygen through methane at 0° C and 1 atm is 0.184 cm ² /s.	4
4B.	Explain the following – (i) Penetration theory (ii) HETP (iii) Schmidt Number	3
4C.	Solubility of CO_2 in PET polymer is 0.73 cm ³ (STP)/[cm ³ (polymer).atm]. A sachet of PET film of 25 µm thickness containing food product is packed under positive pressure of CO ₂ at 38 mm Hg. Compute the rate of diffusion of CO ₂ in g(O ₂)/m ² .d. Diffusivity of CO ₂ through PET is 2x10 ⁻⁹ cm ² /s.	3
5A.	What are the desirable characteristics of plastic films for food packaging	2
5B.	Mass transfer coefficient of O ₂ from a bubble rising in water is calculated using Winnikow equation (1967) at 25 ^o C. Sh = $\frac{2}{\sqrt{\pi}} [1 - \frac{2.89}{\sqrt{Re}}]^{0.5} \sqrt{Pe}$ Compute the flux of oxygen from rising air bubbles (g-O ₂ /m ² .s). Data: Average bubble diameter =3 mm; Rise velocity = 5cm/s; Viscosity of water = 8.9x10 ⁻⁴ Pa.s; density of water=995 kg/m ³ ; Diffusivity of oxygen in water =2x10 ⁻⁵ cm ² /s; Saturation Oxygen concentration in water = 8 ppm (by wt, g/g), Concentration of oxygen in bulk =4 ppm (by wt, g/g).	4
5C.	 During absorption of CCl₄ from a mixture of air-CCl₄ by an organic oil, the individual gas and liquid phase mass transfer have been estimated to be 0.32 and 5.26 kmol/[hr. m². mol fraction] respectively. At a given location, the mole fractions of CCl₄ is found to be 0.3 and 0.01 in gas and liquid respectively. The equilibrium relationship under operating condition is given by y = 20*x, where x and y are mole fractions of CCl₄ in gas and liquid phases respectively. (i) Estimate overall mass transfer coefficients K_x and K_y in kmol/[hr. m². mol fraction] (ii) Estimate the flux of CCl₄ in kmol/(hr. m²) (iii) Determine interfacial composition 	4