Question Paper

Exam Date & Time: 25-Nov-2019 (08:30 AM - 11:30 AM)

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FIRST SEMESTER B.TECH END SEMESTER EXAMINATIONS, NOV 2019 Engineering Physics [PHY 1051 - 2018 -PHY]

Duration: 180 mins.

Marks: 50

Α

Answer all the questions.

Instructions to Candidates: Answer ALL questions Missing data may be suitably assumed

- What are Newton rings? Why they are circular? With a neat geometry, obtain an expression for the radii of dark rings. Mention two applications of Newton rings experiment.
 A)
 B) The first-order diffraction maximum is observed at 12.6° for a crystal having a spacing between plane (3) of atoms of 0.250 nm. (a) What wavelength X-ray is used to observe this first-order pattern? (b) At what angle, second order maximum is observed (c) How many orders can be observed for this crystal at this wavelength?
 - C) Monochromatic light of wavelength 538 nm falls on a single slit with width 0.10 mm. Consider a point (2) on the screen at 9.0° from the central maximum. Calculate the ratio of the intensity at this point to the intensity at the central maximum.
- 2) Explain Compton Effect. Derive the Compton shift equation. For what value of photon scattering (5) angle, we obtain maximum Compton shift
 - B) Explain the polarization phenomena by a) reflection b) double refraction
- (3)
- C) An electron has a kinetic energy of 3.0 eV. (a) Find its wavelength. (b) Also find the wavelength of a (2) photon having the same energy. Mass of an electron is 9.1×10^{-31} Kg ; speed of light in vacuum = 3×10^8 m/s; Planck's constant= 6.63×10^{-34} Js; Avagadro number = 6.023×10^{23} / mol; Boltzmann constant= 1.38×10^{-23} J/K
- 3) Sketch the potential-well diagram of finite height U and length L, obtain the general solution of the (5) Schrödinger equation for a particle of mass m in it.
 - A) B)

A)

- A quantum simple harmonic oscillator consists of an electron bound by a restoring force proportional (3) to its position relative to a certain equilibrium point. The proportionality constant is 8.99 N/m. Calculate its energy in level n = 3. What is the longest wavelength of light that can excite the oscillator? Mass of an electron is 9.1×10^{-31} Kg ; speed of light in vacuum = 3×10^8 m/s; Planck's constant= 6.63×10^{-34} Js; Avagadro number = 6.023×10^{23} / mol; Boltzmann constant= 1.38×10^{-23} J/K
- C) A tungsten target is struck by electrons that have been accelerated from rest through a 40.0-kV (2) potential difference. Find the shortest wavelength of the radiation emitted. What is the kinetic energy of the accelerating electrons? Mass of an electron is 9.1×10^{-31} Kg ; speed of light in vacuum = 3×10^{8} m/s; Planck's constant= 6.63×10^{-34} Js; Avagadro number = 6.023×10^{23} / mol; Boltzmann constant= 1.38×10^{-23} J/K
- 4) Explain the following terms: a) population inversion b) meta stable states. With relevant diagrams, (5) briefly explain the working principle of laser.

A)

- B) Explain briefly (a) ionic bonding, (b) covalent bonding, (c) van der Walls bonding
- C) For copper at 300 K, calculate the probability that a state with an energy equal to 99.0% of the Fermi (2) energy is occupied. Fermi energy of copper is 7.05 eV. Mass of an electron is $9.1 \times 10^{-31} \text{ Kg}$; speed of light in vacuum = $3 \times 10^8 \text{ m/s}$; Planck's constant= $6.63 \times 10^{-34} \text{ Js}$; Avagadro number = $6.023 \times 10^{23} \text{ / mol}$; Boltzmann constant= $1.38 \times 10^{-23} \text{ JK}$
- 5) Obtain an expression for rotational energy of a diatomic molecule. Sketch schematically these (5) rotational energy levels.

A)

- B) Sodium is a monovalent metal having a density of 971 kg/m³ and a molar mass of 0.023 kg/mol. Use (3) this information to calculate (a) the density of charge carriers and (b) the Fermi energy. Mass of an electron is 9.1x10-³¹ Kg ; speed of light in vacuum = 3x10^s m/s; Planck's constant=6.63x10-³⁴ Js; Avagadro number = 6.023x10²³ / mol; Boltzmann constant=1.38x10-²³ J/K
- C) A light-emitting diode (LED) made of the semiconductor GaAsP emits red light (λ = 650nm). (2) Determine the energy-band gap E_g in the semiconductor. Calculate the frequency of the emitted photon. Mass of an electron is 9.1x10⁻³¹ Kg ; speed of light in vacuum = 3x10⁸ m/s; Planck's constant=6.63x10⁻³⁴ Js; Avagadro number = 6.023x10²³ / mol; Boltzmann constant=1.38x10-23 J/K

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