

Reg. No.



# MANIPAL INSTITUTE OF TECHNOLOGY

## MANIPAL

(A constituent unit of MAHE, Manipal)

### V SEMESTER B.TECH. ONLINE PROCTORED END SEMESTER

### EXAMINATIONS DEC 2021

### SUBJECT: MASS TRANSFER II [CHE 3152]

### REVISED CREDIT SYSTEM

### (23/12/2021)

Time: 75 minutes

MAX. MARKS: 20

#### Instructions to Candidates:

- ❖ Answer **ALL** the questions.
- ❖ Missing data may be suitable assumed.

1A.	<p>A solution of carbon tetrachloride and carbon disulfide containing 50 wt% each is to be continuously fractionated at standard atmospheric pressure at the rate of 5500kg/h. The distillate product is to contain 92 wt % carbon disulfide, the residue 0.8 wt %. The feed will be 40 mol% vaporized before it enters the tower. A total condenser will be used, and the reflux will be returned at the bubble point. The equilibrium data (<math>x, y^*</math> = mole fraction <math>CS_2</math>) is as follows:</p> <p>Molecular weight of carbon disulfide and carbon tetrachloride are 76 g/mol and 154 g/mol respectively.</p> <table border="1" data-bbox="632 1272 978 1771"> <thead> <tr> <th>T (°C)</th><th>x</th><th>y*</th></tr> </thead> <tbody> <tr><td>76.7</td><td>0</td><td>0</td></tr> <tr><td>74.9</td><td>0.0296</td><td>0.0823</td></tr> <tr><td>73.1</td><td>0.0615</td><td>0.1555</td></tr> <tr><td>70.3</td><td>0.1106</td><td>0.2660</td></tr> <tr><td>68.6</td><td>0.1435</td><td>0.3325</td></tr> <tr><td>63.8</td><td>0.2585</td><td>0.4950</td></tr> <tr><td>59.3</td><td>0.3908</td><td>0.6340</td></tr> <tr><td>55.3</td><td>0.5318</td><td>0.7470</td></tr> <tr><td>52.3</td><td>0.6630</td><td>0.8290</td></tr> <tr><td>50.4</td><td>0.7574</td><td>0.8780</td></tr> <tr><td>48.5</td><td>0.8604</td><td>0.9320</td></tr> <tr><td>46.3</td><td>1</td><td>1</td></tr> </tbody> </table> <p>Determine the number of theoretical trays required at a reflux ratio equal to the twice the minimum.</p>	T (°C)	x	y*	76.7	0	0	74.9	0.0296	0.0823	73.1	0.0615	0.1555	70.3	0.1106	0.2660	68.6	0.1435	0.3325	63.8	0.2585	0.4950	59.3	0.3908	0.6340	55.3	0.5318	0.7470	52.3	0.6630	0.8290	50.4	0.7574	0.8780	48.5	0.8604	0.9320	46.3	1	1	05
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1B.	<p>Dilute ethanol-water solutions can be continuously rectified to give at best the mixtures containing 89.4 mole % ethanol at atmospheric pressure, since this is the composition of minimum boiling azeotrope in the binary system. Ethanol can be further purified either by using n-pentane as entrainer or ethylene glycol as solvent. Write short notes on the methods which uses the above-mentioned compounds in the purification of ethanol and comment on the most desirable method.</p>	02																																							

1C.	Discuss about spiral wound membranes and bundle of hollow fibers with schematic representation. Also, which among these is generally used in RO systems	03																																																																														
2A.	900 kg of crushed oil seeds (22% oil, 78% meal) is extracted in a three-stage cross-current unit using 600 kg of pure hexane in each stage. The equilibrium data are as follows:	05																																																																														
	<table><tr><th colspan="3">Overflow (100 kg) solution</th><th colspan="3">Underflow (100 kg) slurry</th></tr><tr><th>W<sub>A</sub> (kg)</th><th>W<sub>B</sub> (kg)</th><th>W<sub>C</sub> (kg)</th><th>W'<sub>A</sub> (kg)</th><th>W'<sub>B</sub> (kg)</th><th>W'<sub>C</sub> (kg)</th></tr><tr><td>0.3</td><td>99.7</td><td>0</td><td>67.2</td><td>32.8</td><td>0</td></tr><tr><td>0.45</td><td>90.6</td><td>8.95</td><td>67.1</td><td>29.94</td><td>2.96</td></tr><tr><td>0.54</td><td>84.54</td><td>14.92</td><td>66.93</td><td>28.11</td><td>4.96</td></tr><tr><td>0.70</td><td>74.47</td><td>24.83</td><td>66.58</td><td>25.06</td><td>8.36</td></tr><tr><td>0.77</td><td>69.46</td><td>29.77</td><td>66.26</td><td>23.62</td><td>10.12</td></tr><tr><td>0.91</td><td>60.44</td><td>38.65</td><td>65.75</td><td>20.9</td><td>13.35</td></tr><tr><td>0.99</td><td>54.45</td><td>44.56</td><td>65.33</td><td>19.07</td><td>15.6</td></tr><tr><td>1.19</td><td>44.46</td><td>54.35</td><td>64.39</td><td>16.02</td><td>19.59</td></tr><tr><td>1.28</td><td>38.50</td><td>60.22</td><td>63.77</td><td>14.13</td><td>22.10</td></tr><tr><td>1.28</td><td>34.55</td><td>64.17</td><td>63.23</td><td>12.87</td><td>23.90</td></tr><tr><td>1.48</td><td>24.63</td><td>73.89</td><td>61.54</td><td>9.61</td><td>28.85</td></tr></table>		Overflow (100 kg) solution			Underflow (100 kg) slurry			W <sub>A</sub> (kg)	W <sub>B</sub> (kg)	W <sub>C</sub> (kg)	W' <sub>A</sub> (kg)	W' <sub>B</sub> (kg)	W' <sub>C</sub> (kg)	0.3	99.7	0	67.2	32.8	0	0.45	90.6	8.95	67.1	29.94	2.96	0.54	84.54	14.92	66.93	28.11	4.96	0.70	74.47	24.83	66.58	25.06	8.36	0.77	69.46	29.77	66.26	23.62	10.12	0.91	60.44	38.65	65.75	20.9	13.35	0.99	54.45	44.56	65.33	19.07	15.6	1.19	44.46	54.35	64.39	16.02	19.59	1.28	38.50	60.22	63.77	14.13	22.10	1.28	34.55	64.17	63.23	12.87	23.90	1.48	24.63	73.89	61.54	9.61	28.85
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Calculate the fraction of oil extracted in a three-stage cross-current unit using PS method.																																																																																
2B.	Consider question 2A. Calculate the fraction of oil extracted in a single stage contactor for the same volume (1800kg) of the solvent and comment on the result.	03																																																																														
2C.	Write a short note on any one solid-liquid contacting equipment which does not result in the clogging by fines with neat schematic diagram.	02																																																																														

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