

Time: 2 Hours

DEPARTMENT OF SCIENCES, IV SEMESTER M. Sc. (CHEMISTRY) END SEMESTER EXAMINATIONS, JUNE 2021

Chemistry of Natural Products [CHM 5013]

(REVISED CREDIT SYSTEM-2017) Date:12-06-2021

MAX. MARKS: 40

N	Note: (i) Answer ANY 4 questions	
	(ii) Draw diagrams, and write equations wherever necessary	
1A.	Write the difference between Hoffmann and von-Brown degradation reaction	3
B.	What are the structural variations that you observe in isomers of Viţamin A? How do you differentiate them by chemical tests?	3
С	Propose the structure of conhydrine, an alkaloid with empirical formula C ₈ H ₁₇ NO. This molecules consumes (a) 2 equivalents of acetyl chloride (b) 1 equivalent of MeI (c) oxidized product consumes 1 equivalent of 2,4-DNP (d) basic skeleton resembles pyridine	4
2A.	What are the tests used to detect the different types of the carbonyl group in terpenoids? Explain with example.	3
В	Write the chemical tests to distinguish primary, secondary and tertiary amines	3
C.	Write the general reaction for the preparation of flavones and isoflavones	4
3A.	Write Emde degradation product for the following molecules.	3
В.	Describe the Baker-Venkatraman synthesis. Explain why this method cannot be employed for the synthesis of isoflavonoids.	3
C.	Identify the functional groups present in the plant extract having one N and three O atoms, from the following chemical reactions: (a) Benzoylation gives mono derivative (b) Dibromo derivative results from Br ₂ solution (c) Boiling with HCl eliminates CH ₃ Cl (d) Gives effervescence with NaHCO ₃	4

4A. How do you analyze isolated and conjugated C=C bond qualitatively and quantitatively in an organic compound?

3

3

4

3

B. Menthol (C₁₀H₂₀O) and menthone (C₁₀H₁₈O) are having similar structure. Propose chemical reactions to differentiate these two molecules. Justify whether inter-conversion is possible between menthol and menthone.

C. Write the product formed when the following molecules react with (a) ethane (b) ozone and (c) acetyl chloride

- 5A. Explain your reasoning.
 - (i) testosterone and androsterone are difficult to differentiate using chemical methods,
 - (ii) esterdiol to estertriol conversion is difficult
 - (iii) Carotenoids exhibit a large number of geometrical isomers and colour in the visible range
 - B. Explain the general method of isolating terpenoids from plant source.
 - C. Identify A to D from the following data;
 Molecule A with empirical formula C₅H₈NOCl is treated with B to produce C. This compound further reacts with 2,4 DNP. Compound A consumes two equivalents of acetyl chloride and it can also exhibit keto-enol tautomerism. It also reacts with ethene to give different product D. The compounds A, C and D give a broad peak at 3400 cm⁻¹ in the IR.
- 6A. Explain the nature of heteroatom present in the terpenoid (C₁₁H₂₂O) which does not react with acetyl chloride.
- B. How do you differentiate the following functional groups present in alkaloids? 3

 (i) Amide and ester

 (ii) R-NH₂ and R₂NH

 (iii) NCH₃ and OCH₃
- C. Write the difference between (i) epiphasic and hypophasic carotenoids 4 (ii) xanthophylls and carotenes.