

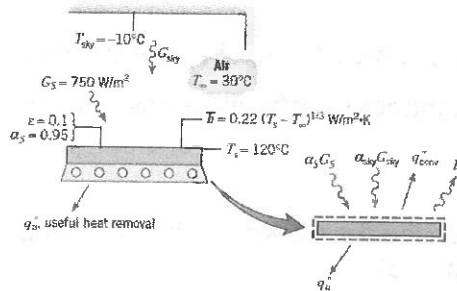
HTO - CHE - 2253

Q.P.

IV Sem

Type: DES

Q1. A flat-plate solar collector with no cover plate has a selective absorber surface of emissivity 0.1 and solar absorptivity 0.95. At a given time of day the absorber surface temperature  $T_s$  is  $120^\circ\text{C}$  when the solar irradiation is  $750\text{ W/m}^2$ , the effective sky temperature is  $-10^\circ\text{C}$ , and the ambient air temperature  $T_\infty$  is  $30^\circ\text{C}$ . Assume that the heat transfer convection coefficient for the calm day conditions can be estimated from  $\bar{h} = 0.22(T_s - T_\infty)^{1/3}\text{ W/m}^2 \cdot \text{K}$ .



Calculate the useful heat removal rate ( $\text{W/m}^2$ ) from the collector for these conditions. What is the corresponding efficiency of the collector? Consider  $\alpha_{\text{sky}} = 0.1$  (4)

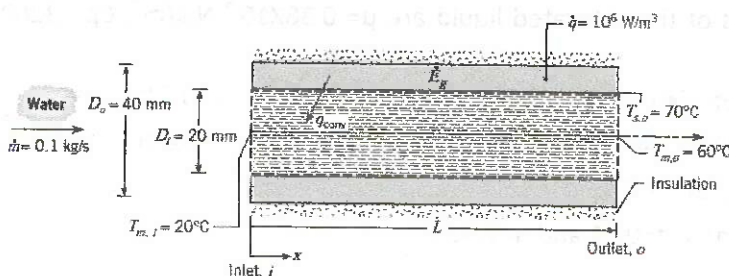
✓ Q2. A glass window's inner and outer surface temperatures of 5 mm thick are  $15^\circ\text{C}$  and  $5^\circ\text{C}$ . What is the heat loss through a window that is 1 m by 3 m on a side? The thermal conductivity of glass is  $1.4\text{ W/m} \cdot \text{K}$ . Define Conduction, convection, and radiation. (3)

Q3. Derive the Equation that is used to compute the temperature reached by the solid at some time  $t$ . (3)

Q4.

A system for heating water from an inlet temperature of  $T_{m,i} = 20^\circ\text{C}$  to an outlet temperature of  $T_{m,o} = 60^\circ\text{C}$  involves passing the water through a thick-walled tube having inner and outer diameters of 20 and 40 mm. The outer surface of the tube is well insulated, and electrical heating within the wall provides for a uniform generation rate of  $\dot{q} = 10^6\text{ W/m}^3$ .

1. For a water mass flow rate of  $\dot{m} = 0.1\text{ kg/s}$ , how long must the tube be to achieve the desired outlet temperature?
2. If the inner surface temperature of the tube is  $T_s = 70^\circ\text{C}$  at the outlet, what is the local convection heat transfer coefficient at the outlet?



Consider the  $C_p$  for water as  $4179 \text{ J/kg.K}$ . (5)

Q5. Define Radiosity, the total hemispherical emissive power and spectral, directional emissivity. (3)

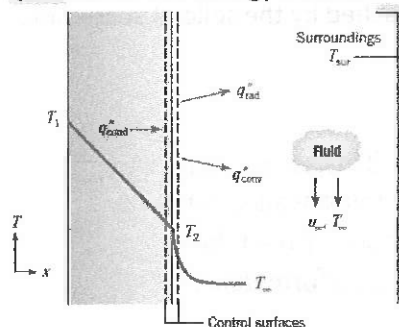
Q6. Explain any two ways to increase the heat transfer coefficient. (2)

Q7. The engine oil at  $150^\circ\text{C}$  is cooled to  $80^\circ\text{C}$  in a parallel flow heat exchanger by water entering at  $25^\circ\text{C}$  and leaving at  $60^\circ\text{C}$ . Estimate the exchanger's effectiveness and the number of transfer units. (4)

Q8. Define the Capacity and economy of evaporators. Describe the working of the short-tube and falling film evaporators with the help of a neat diagram. (3)

Q9. Consider a plane composite wall that is composed of two materials of thermal conductivities  $k_A = 0.1 \text{ W/m.K}$  and  $k_B = 0.04 \text{ W/m.K}$  and thicknesses  $L_A = 10 \text{ mm}$  and  $L_B = 20 \text{ mm}$ . The contact resistance at the interface between the two materials is known to be  $0.30 \text{ m}^2 \cdot \text{K/W}$ . Material A adjoins a fluid at  $200^\circ\text{C}$  for which  $h = 10 \text{ W/m}^2 \cdot \text{K}$  and material B adjoins a fluid at  $40^\circ\text{C}$  for which  $h = 20 \text{ W/m}^2 \cdot \text{K}$ . What is the rate of heat transfer through a wall that is  $2 \text{ m}$  high by  $2.5 \text{ m}$  wide? (3)

Q10. Write the energy balance for the system shown in the diagram.



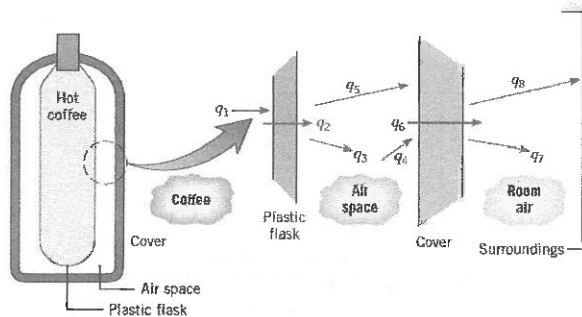
Define Absorptivity and irradiation. (2)

Q11. Saturated ethylene glycol at  $1 \text{ atm}$  is heated by a horizontal chromium-plated surface which has a diameter of  $200 \text{ mm}$  and is maintained at  $480 \text{ K}$ . Estimate the heating power requirement and the rate of evaporation. The properties of the saturated liquid are  $\mu = 0.38 \times 10^{-3} \text{ N.s/m}^2$ ,  $C_p = 3280 \text{ J/Kg.K}$ , and  $\text{Pr} = 8.7$ .

**PROPERTIES:** Table A-5, Saturated ethylene glycol ( $1 \text{ atm}$ ):  $T_{\text{sat}} = 470 \text{ K}$ ,  $h_{fg} = 812 \text{ kJ/kg}$ ,  $\rho_f = 1111 \text{ kg/m}^3$ ,  $\sigma = 32.7 \times 10^{-3} \text{ N/m}$ ; Saturated ethylene glycol (given,  $470 \text{ K}$ ):  $\rho_v = 1.66 \text{ kg/m}^3$ ,  $\mu_\ell = 0.38 \times 10^{-3} \text{ N.s/m}^2$ ,  $c_{p,\ell} = 3280 \text{ J/kg.K}$ ,  $\text{Pr}_\ell = 8.7$ .

The saturated vapor density is  $1.66 \text{ kg/m}^3$ .  $C_{s,f} = 0.01$  and  $n = 1$ . (5)

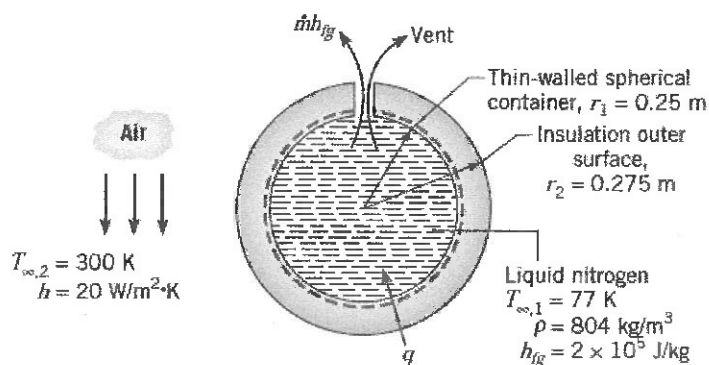
Q12. A closed container filled with hot coffee is in a room whose air and walls are at a fixed temperature. Identify all heat transfer processes that contribute to the cooling of the coffee.



(3)

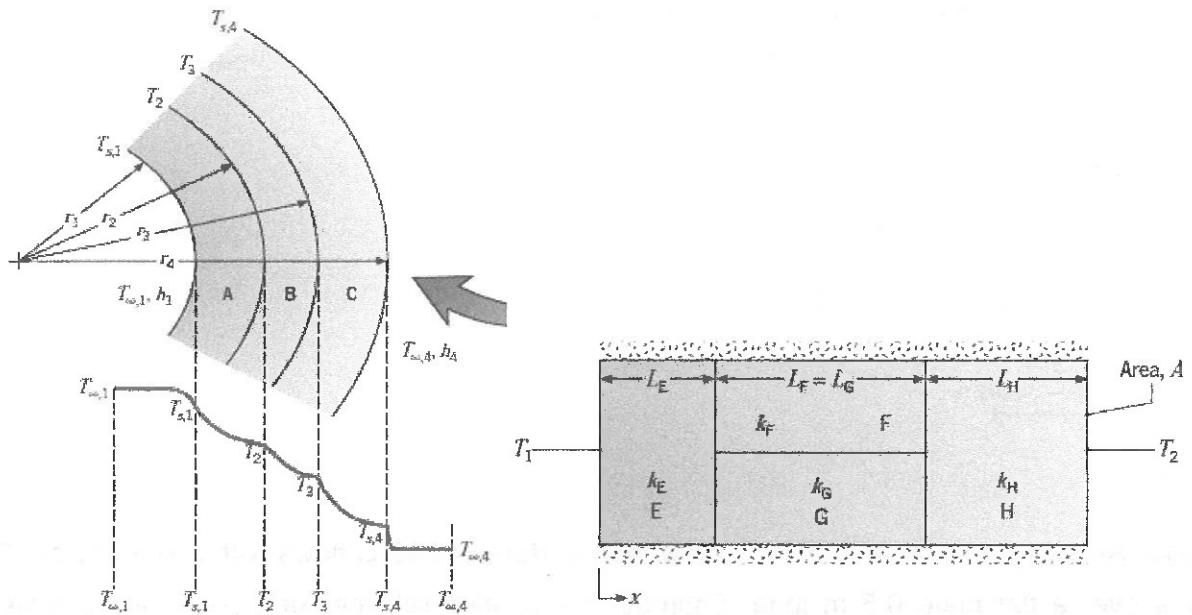
Q13. Air at a pressure of  $6 \text{ kN/m}^2$  and a temperature of  $300^\circ\text{C}$  flows with a velocity of  $10 \text{ m/s}$  over a flat plate  $0.5 \text{ m}$  long. Estimate the cooling rate per unit width of the plate needed to maintain it at a surface temperature of  $27^\circ\text{C}$ .  $Pr=0.687$ ,  $\nu=5.21 \times 10^{-4} \text{ m}^2/\text{s}$ ,  $k=36.4 \times 10^{-3} \text{ W/m}\cdot\text{K}$  (3)

Q14. A spherical, thin-walled metallic container is used to store liquid nitrogen at  $77 \text{ K}$ . The container has a diameter of  $0.5 \text{ m}$  and is covered with evacuated, reflective insulation composed of silica powder. The insulation is  $25 \text{ mm}$  thick, and its outer surface is exposed to ambient air at  $300 \text{ K}$ . The convection coefficient is known to be  $20 \text{ W/m}^2\cdot\text{K}$ . The latent heat of vaporization and the density of liquid nitrogen are  $2 \times 10^5 \text{ J/kg}$  and  $804 \text{ kg/m}^3$ , respectively.



What is the rate of heat transfer to the liquid nitrogen? What is the rate of liquid boil-off per day? (4)

Q15. For the composite systems shown in the diagram, express the heat transfer rate for the cylindrical system and draw the thermal circuits (series and parallel) for the composite wall.



(3)