Exam Date & Time: 08-Jan-2024 (09:30 AM - 12:30 PM)



## MANIPAL ACADEMY OF HIGHER EDUCATION

THIRD SEMESTER B.TECH END SEMESTER MAKE UP EXAMINATIONS, JAN 2024

PHYSICAL AND ORGANIC CHEMISTRY [CHM 2121]

Ma	rks: 50	Duration: 180	mins
		A	
An	swer all t	he questions.	
Ins	tructions	to Candidates: Answer ALL questions Missing data may be suitably assumed	
1)		i) Derive an expression for the determination of rate constant of first order reaction.	
		ii) Explain the solution properties of non ideal system showing negative deviation from Raoult's law.	(4)
	A)		
	B)	Derive an equation for the following.	
		i) The relationship between the free energy of a solvent and the free energy of a nonvolatile solute in solution.	(4)
		ii) The relationship between the elevation in boiling point to the mole fraction of the solute.	
	C)	Differentiate between the following.	
	C)		
		i) Physisorption and chemisorption.	(2)
		ii) Order and molecularity.	
2)		i) Explain the phase diagram of the water system.	
		ii) Explain the potentiometric titration of weak acid against standard sodium hydroxide solution.	(4)
	A)		
	B)	i) Derive an equation to show the relationship between the vapor pressure of small liquid drops and the drop size.	vav
		ii) Explain the phase diagram of two-component system forming a Eutectic mixture.	(4)
	C)	Explain the solubility of the nicotine-water system with temperature.	(2)
3)		i) Explain the boiling temperature- composition curve for the chloroform-acetone system.	
		ii) Explain the adsorption of solute on charcoal in solution.	(4)
	A)		
	B)	Discuss the resonance structure of benzene. Explain the electrophilic substitution reaction of benzene with the suitable mechanism.	(4)
	C)	Explain the Huckel's rule of Aromaticity with an example.	(2)
4)		Differentiate between bathochromic and hypsochromic shift. How dyes are classified based on application and structure?	(4)
	A)		
	B)	Explain the classification of carbohydrates. Discuss the structure of Glucose.	

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C) What are heterocyclic compounds? Why pyridine is more basic than pyrrole?

(2)

 Define carbocations. Discuss the stability of carbocations based on the inductive effect and resonance effect.

(4)

- A)
- B) Assign and explain the suitable nomenclature for the following compounds.

$$C = C$$
 $C_2H_5$ 
 $F_L$ 
 $C = C$ 
 $C_2H_5$ 
 $C_1$ 
 $C_2$ 
 $C_2$ 
 $C_3$ 
 $C_4$ 
 $C_4$ 
 $C_4$ 
 $C_5$ 
 $C_5$ 
 $C_7$ 
 $C_8$ 
 $C_8$ 

C) Mention any four factors that affect the strength of acids and bases.

(2)

----End----

11.1